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Anticancer Activity and Green Synthesized TiO₂ Nanoparticles From Calotropis Gigantea Leaves Extract

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Abstract: *Aim & Objective: Research the Anticancer Abilities of TiO₂ Nanoparticles Obtain using Calotropis Gigantea by liver cells (HepG2). Materials & Methods: Various plant materials for the synthesis of nanoparticles are considered in green technology. In this present study, reports the synthesis of Tio₂ nanoparticles that were synthesized from Calotropis gigantea leaf, and the formation of Tio₂ nanoparticles was observed with different time intervals. The characterization was recorded from scanning electron microscope, X-ray diffraction, Fourier transfer infrared spectra, DLS, UV-Vis spectrum. The supporting green synthesis and characterization of Tio₂ nanoparticles. Result: The result is recorded from MTT assay and ETBR/AO Staining.*

Keywords: *Calotropis gigantea, Tio₂ Nanoparticles, MTT, ETBR/AO.*

I. INTRODUCTION

Cancer are most serious health problems globally. Liver cancer is one of the maximum competitive occurrences of most cancers with a excessive fatality rate global huge[1]. Hepatocellular carcinoma (HCC) is the maximum common form of number one liver cancer, with cholangiocarcinoma accounting for the remainder. *C.gigantea* (Apocynaceae, Asclepedaceae) is usually used globally as a traditional medication for the treatment of several elements. *C.gigantea* plant is extensively grown in many countries in Africa, eastern Asia, and South-east Asia. Extracts from all elements of this plant had been shown to have an expansion of organic activities[2]. The findings of this studies may provide valuable statistics for the improvement of low-chance most cancers remedies based on plant extracts. The blended healing benefits of *C.gigantea* leaves extracts may be beneficial for destiny anticancer treatment regimens.



Plant of Calotropis Gigantea

A. Plant Profile

Plant name: Calotropis gigantea

Family name: Apocynaceae

Synonyms: Calotropis procera, Calotropis acia

Common name: Giant milkweed, Crown flower

Aakh Parts used: leaves.

B. Classification

Kingdom: Plantea

Subkingdom: Tracheobionata
Superdivision: Spermatophyta
Division: Magnoliopsida
Subclass: Asteridea
Order: Gentianales
Family: Apocynacea
Subfamily: Asclepiadoideae
Genus: Calotropis
Species: gigantea, procera, acia

II. MATERIALS AND METHODS

A. Collection and Authentication of Plant

Calotropis gigantea was collected from Mallasamudram, Namakkal district, TamilNadu, India in the month of August, 2023. The plant material was identified and authenticated by Dr. D.Vijayakumar Assistant Professor, Mahendra Institute of Technology Namakkal, TamilNadu, India.

B. Preparation of Plant Extract

Calotropis gigantea have been washed 3 times with faucet water then double distilled water and dried for 10 days via using a hot air oven at 40°C. The dried leaf had been grinded right into a excellent powder[3]. In 100ml of conical flasks, 10 grams of leaf powder were mixed with 100 mL of distilled water. The aggregate become boiled at 200°C for 20 min. The extract turned into filtered via muslin cloth. The filtered extract become stored at 4°C within the refrigerator for further use.

C. Bulk Synthesis of TiO_2 NPs Using *Calotropis gigantea* leaf

10ml of *calotropis gigantea* leaf extract was mixed with TiO_2 Solution (90ml) in dark condition for 24hours. A change in the color of the solution from white to light yellow indicated the synthesis of TiO_2 .

D. Characterization of synthesized TiO_2 Nanoparticles

The biosynthesized TiO_2 NPs have been characterized consistent with the technique. The form and length of TiO_2 nanoparticles were determined by using Scanning electron microscopy (SEM)[4]. X-ray diffraction (XRD) is the main method for crystallographic characterization for bulk, nano and thin film materials[5]. Fourier transform infrared spectroscopy (FTIR) turned into used to signify the adjustments and the composition at the floor of the synthesized nanoparticles. The particle size have been measured by means of dynamic light scattering (DLS) the usage of a Zetasizer Nano ZS90[6]. UV-via spectra were measured using a UV-2450.

E. MTT Assay

- 1) **Principle:** MTT (3-(4, 5 dimethylthiazol-2-yl)-2, 5-diphenyl tetrazolium bromide) assay, is based totally on the capability of a mitochondrial dehydrogenase enzyme of viable cells to cleave the tetrazolium rings of the light yellow MTT and form a dark blue colored formazan crystal which is essentially impermeable to cell membranes, therefore resulting in its accumulation within healthy cells[7]. Solubilization of cells with the addition of detergents (DMSO) results in the liberation of crystals which might be solubilized. The number of surviving cells is directly proportional to the level of formazan product created[8]. The color may be quantified using a multi-well plate reader.
- 2) **Materials Required:** Fetal Bovine Serum (FBS) and antibiotic solution have been from Gibco (USA), DMSO (Dimethyl sulfoxide) and MTT (3-(4,5 dimethylthiazol-2-yl)-2,5-diphenyl tetrazolium bromide) (5 mg/ml) were from Sigma, (USA), DMEM medium, 1X PBS, (India)[9]. 96 well tissue culture plate and wash beaker have been from Tarson (India)[10].
- 3) **Procedure**
 - a) **Cell Culture:** Change liver cellular line have been purchased from NCCS, Pune and have been cultured in liquid medium (DMEM) supplemented 10% Fetal Bovine Serum (FBS), 100 µg/ml penicillin and 100 µg/ml streptomycin, and maintained under an atmosphere of 5% CO_2 at 37°C.
 - b) **MTT assay:** The Test sample was tested for *in vitro* cytotoxicity, using Change liver cells by MTT assay. Briefly, the cultured Change liver cells were harvested by trypsinization and pooled in a 15 ml tube[11].

Then, the cells were plated at a density of 1×10^5 cells/ml cells/well (200 μ L) into the 96-well tissue culture plate in DMEM medium containing 10 % FBS and 1% antibiotic solution for 24-48 hour at 37°C. The wells were washed with sterile PBS and treated with the Test sample in a serum-free DMEM medium[12]. Every sample was replicated three times and the cells were incubated at 37°C in a humidified 5% CO₂ incubator for 24 h. After incubation, MTT (10 μ L of 5 mg/ml) was added to each well and the cells were incubated for another 2-4 h until purple precipitates were clearly visible under an inverted microscope[13]. Later, the medium together with MTT (220 μ L) was aspirated off the wells and washed with 1X PBS (200 μ L). Furthermore, to dissolve formazan crystals, DMSO (100 μ L) was added and the plate changed into shaken for 5 min[14]. The absorbance for each well have measured at 570 nm using a microplate reader (Thermo Fisher Scientific, USA) and IC₅₀ value were calculated using Graph Pad Prism 6.0 software (USA).

$$\text{Formula Cell viability \%} = \text{Test OD/Control OD} \times 100$$

F. ETBR/AO STAINING

1) *Principle:* Fluorescent dyes with aromatic amino or guanidine groups, such as acridine orange (AO), interact with nucleotides to emit fluorescence. EtBr molecules intercalate inside the DNA double helix[15]. AO can molecule intercalate in double-stranded DNA or single-stranded DNA and RNA. One molecule of AO can also interact with one phosphate group of DNA or RNA to form an aggregated, or stacked, structure that emits pink fluorescence with the maximum wavelength at 650 nm. This fluorescent dye is impermeable through the cell membranes of viable cells and can be used as fluorescent indicators of dead cells. Acridine orange is a vital dye and will stain both live and dead cells[16]. Necrotic cells stain orange however have a nuclear morphology such that of viable cells, and not using a condensed chromatin[17]. Ethidium bromide (EtBr) is only taken up by cells when cytoplasmic membrane integrity is lost and stains the nucleus red. EtBr also dominates over AO. For that reason live cells have a normal green nucleus; early apoptotic cells have a bright green nucleus with condensed or fragmented chromatin; late apoptotic cells display condensed and fragmented orange chromatin; cells that have died from direct necrosis have a structurally normal orange nucleus[18]. Ethidium re-emits there energy in the colour of yellow/orange light centered at 590 nm. The fluorescence of ethidium bromide in an aqueous solution is considerably lower than that of the intercalated dye.

2) *Materials required:* DMEM medium, Penicillin/Streptomycin antibiotic solution, Trypsin-EDTA have been purchased from Gibco (USA), EtBr, and Acridine orange also purchased from Sigma Aldrich (USA), Fluorescent Imaging System, (ZOE, Bio-Rad, USA).

3) *Procedure:*

a) *Cell culture*

Change liver cell line have been purchased from NCCS, Pune and was cultured in liquid medium (DMEM) supplemented for 10% Fetal Bovine Serum (FBS), 100 u/ml penicillin, and 100 μ g/ml streptomycin, and maintained under an atmosphere of 5% CO₂ at 37°C[19].

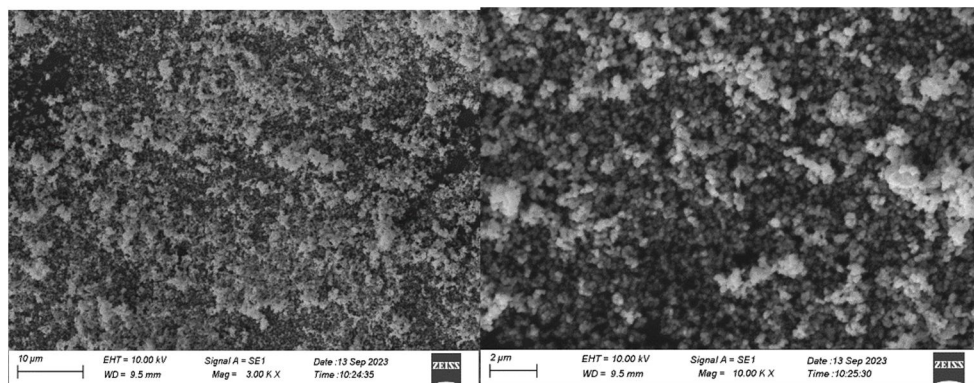
b) *EtBr/AO staining*

Briefly, 5×10^5 cells/ml of Chang liver cells were plated into a 96 well tissue culture plate and incubated for 24 hr in a DMEM growth medium. After incubation, the cells were treated with 44.85 μ g/ml of TiO₂ sample in a serum-free DMEM medium[20]. The plate was incubated at 37°C at a 5% Co₂ incubator for 24 hours[24]. After incubation, 10 μ L of 1 mg/ml acridine orange and ethidium bromide were added to the wells and mixed gently[21]. Finally, the plate was centrifuged at 800 rpm for 2 minutes and evaluated immediately within an hour, and examined at least 100 cells by a Fluorescent Imaging System, (ZOE, Bio-Rad, USA).

III. RESULTS & DISCUSSION

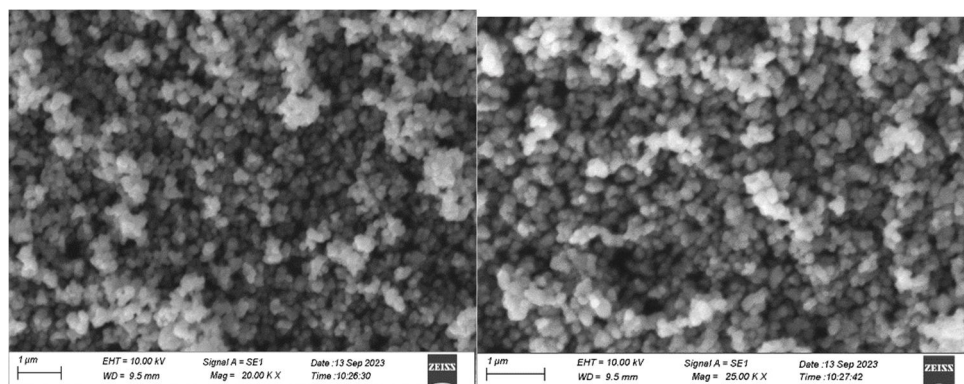
A. *Characterization of synthesized Tio2 Nanoparticles:*

1) *Scanning Electron Microscopy (SEM):* In the SEM Pictures, a microscope is used to qualitatively identify the microstructural developments within the matrix of the stabilized soil specimens[21]. The SEM photos of clays are shown (Unique magnifications); thus, the microstructure is easily located because the pictures can be enlarged[22].



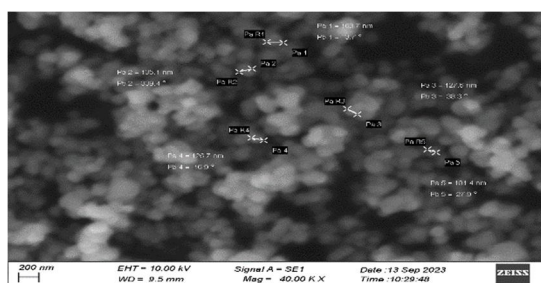
(a) ×3,000 magnification

(b) ×10,000 magnification



(c) ×20,000 magnification

(d) ×25,000 magnification



(e) ×40,000 magnification

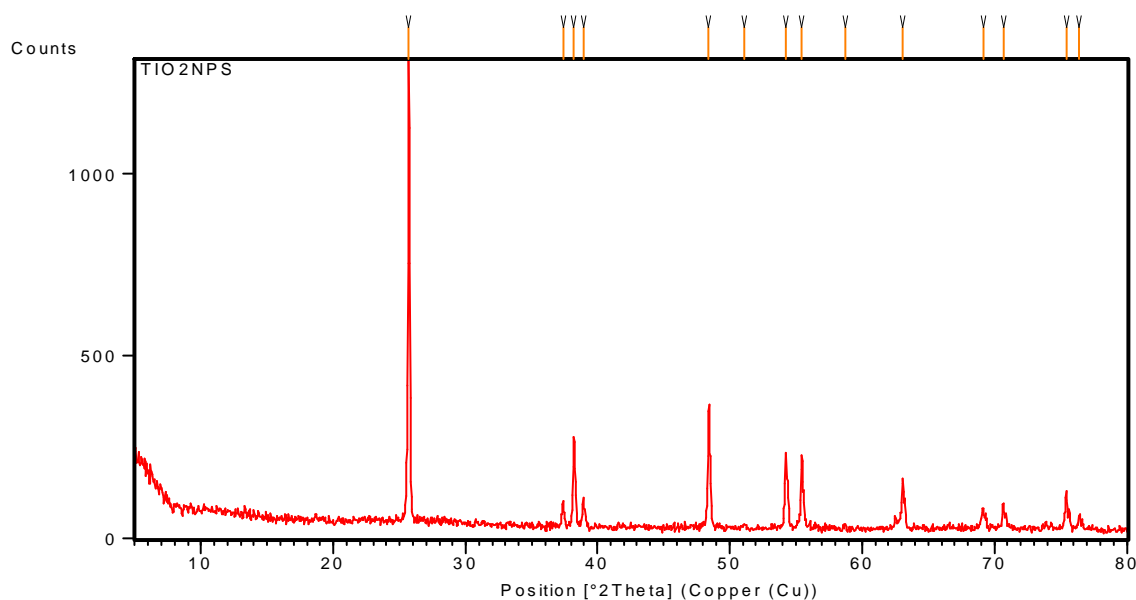
2) X-ray diffraction (XRD):

X-ray diffraction is used most frequently to investigate the structure of biocomposites with embedded nanostructure.

Measurement Conditions:

| | |
|--------------------------|-------------------------|
| Dataset Name | TiO2NPS |
| Raw Data Origin | BRUKER-binary V4 (.RAW) |
| Scan Axis | Gonio |
| Start Position [°2Th.] | 5.0000 |
| End Position [°2Th.] | 80.1600 |
| Step Size [°2Th.] | 0.0400 |
| Scan Step Time [s] | 13.4400 |
| Scan Type | Pre-set time |
| Offset [°2Th.] | 0.0000 |
| Divergence Slit Type | Fixed |
| Divergence Slit Size [°] | 9999.0000 |
| Specimen Length [mm] | 10.00 |

Receiving Slit Size [mm] 0.1000
 Measurement Temperature [°C] 25.00
 Anode Material Cu
 K-Alpha1 [Å] 1.54060
 K-Alpha2 [Å] 1.54443
 K-Beta [Å] 1.39225
 K-A2 / K-A1 Ratio 0.50000
 Generator Settings 30 mA, 40 kV
 Diffractometer Type Theta/Theta
 Diffractometer Number 0
 Goniometer Radius [mm] 240.00
 Dist. Focus-Diverg. Slit [mm] 91.00
 Incident Beam Monochromator No
 Spinning No

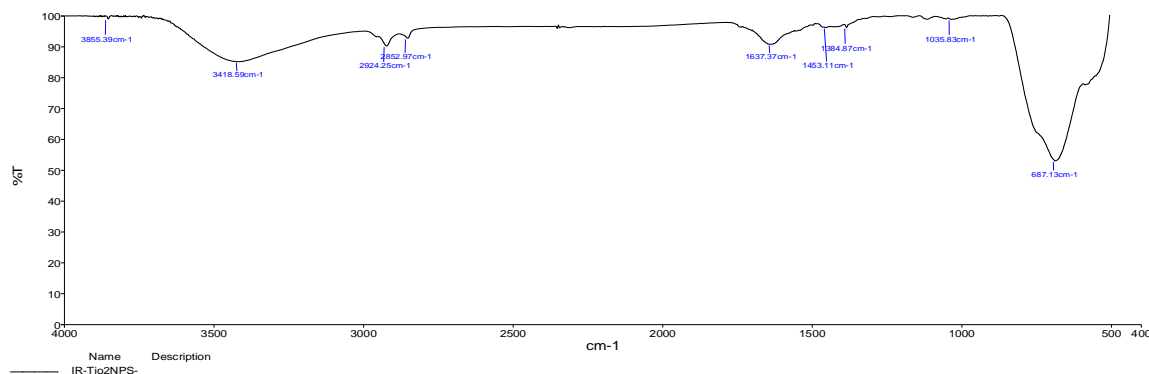


3) Peak List of XRD:

| Pos. [°2Th.] | Height [cts] | FWHM Left [°2Th.] | d-spacing [Å] | Rel. Int. [%] |
|--------------|--------------|-------------------|---------------|---------------|
| 25.7039 | 1180.84 | 0.1181 | 3.46594 | 100.00 |
| 37.3352 | 68.06 | 0.1574 | 2.40859 | 5.76 |
| 38.1740 | 252.27 | 0.1574 | 2.35758 | 21.36 |
| 38.9627 | 79.99 | 0.1968 | 2.31165 | 6.77 |
| 48.4084 | 333.81 | 0.1968 | 1.88038 | 28.27 |
| 51.0752 | 10.33 | 0.4723 | 1.78829 | 0.87 |
| 54.2616 | 219.02 | 0.1574 | 1.69056 | 18.55 |
| 55.4423 | 202.34 | 0.1968 | 1.65733 | 17.14 |
| 58.7508 | 11.95 | 0.2362 | 1.57164 | 1.01 |
| 63.0669 | 139.36 | 0.1181 | 1.47408 | 11.80 |
| 69.1823 | 51.24 | 0.3149 | 1.35796 | 4.34 |
| 70.6650 | 75.06 | 0.1181 | 1.33306 | 6.36 |
| 75.4360 | 107.32 | 0.1181 | 1.26016 | 9.09 |
| 76.4292 | 40.18 | 0.3149 | 1.24624 | 3.40 |

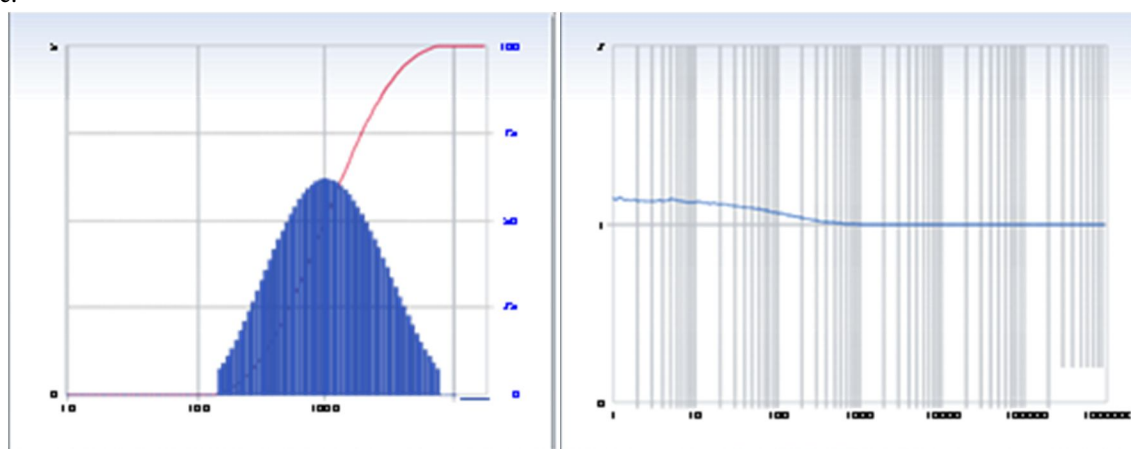
4) Fourier Transform Infrared Spectroscopy (FTIR):

FTIR spectroscopy is a very powerful tool with many applications, however data interpretation is not straightforward[23]. By nature, the total spectrum generated is a series function of absorbed energy response (hence the Fourier Transform portion of the name).



5) DLS (Dynamic Light Scattering):

Dynamic Light Scattering (DLS) analysis allows us to confidently measure the size distribution profiles of particles in the sub-micron range.

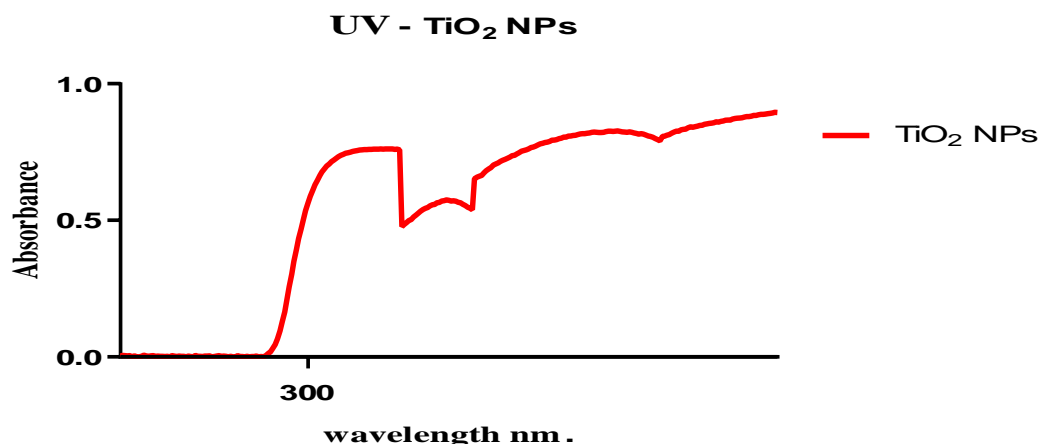


(a) Intensity Distribution

(b) ACF

6) Ultra Violet (UV):

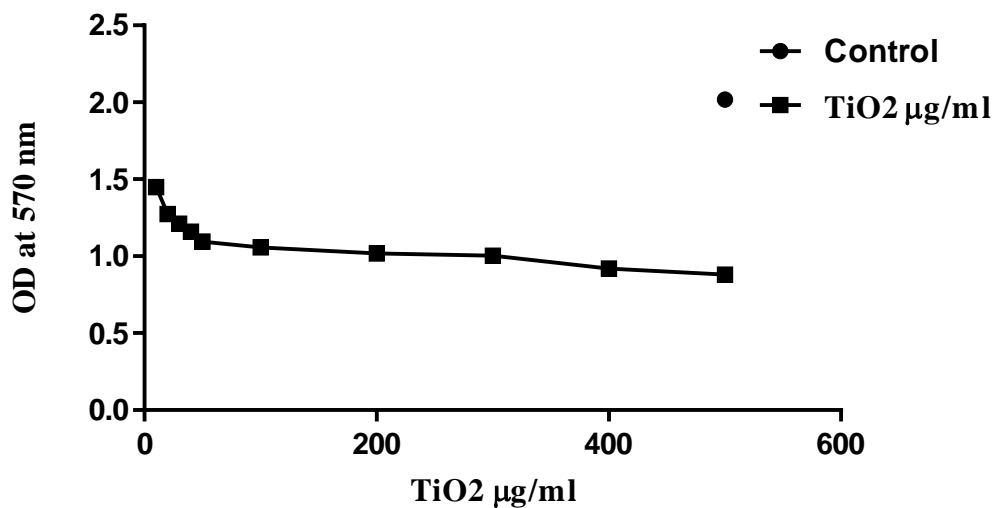
Titanium oxide nanoparticles produced diagnostic UV-Vis peaks at 350-365nm [25].



B. MTT ASSAY IN HEPG2 CELLS

1) OD Value at 570 nm

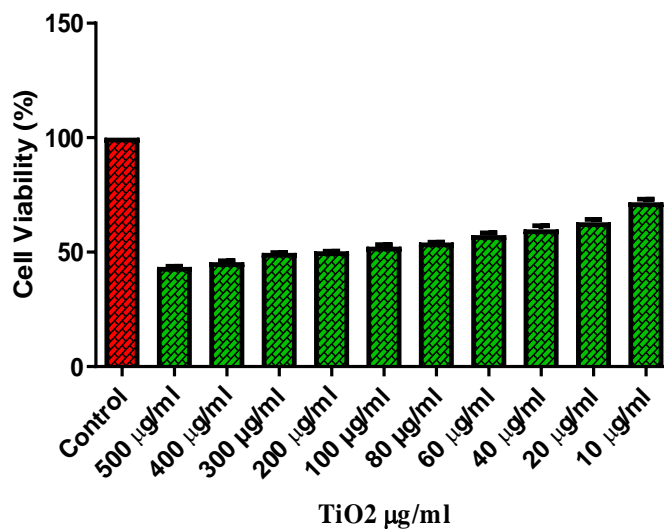
| S. No. | Tested sample concentration (µg/ml) | OD value at 570 nm (in triplicates) | | |
|--------|-------------------------------------|-------------------------------------|-------|-------|
| 1 | Control | 2.012 | 2.019 | 2.021 |
| 2 | 500 µg/ml | 0.876 | 0.877 | 0.887 |
| 3 | 400 µg/ml | 0.901 | 0.929 | 0.929 |
| 4 | 300 µg/ml | 0.998 | 1.004 | 1.007 |
| 5 | 200 µg/ml | 1.013 | 1.017 | 1.02 |
| 6 | 100 µg/ml | 1.043 | 1.051 | 1.08 |
| 7 | 80 µg/ml | 1.091 | 1.094 | 1.1 |
| 8 | 60 µg/ml | 1.133 | 1.171 | 1.173 |
| 9 | 40 µg/ml | 1.176 | 1.216 | 1.242 |
| 10 | 20 µg/ml | 1.244 | 1.289 | 1.289 |
| 11 | 10 µg/ml | 1.421 | 1.441 | 1.482 |



2) Cell Viability (%)

| S. No. | Tested sample concentration (µg/ml) | Cell viability (%) (in triplicates) | | | Mean Value (%) |
|--------|-------------------------------------|-------------------------------------|---------|---------|----------------|
| 1 | Control | 100 | 100 | 100 | 100 |
| 2 | 500 µg/ml | 43.5388 | 43.4373 | 43.8892 | 43.621759 |
| 3 | 400 µg/ml | 44.7813 | 46.0129 | 45.9673 | 45.587178 |
| 4 | 300 µg/ml | 49.6024 | 49.7276 | 49.8268 | 49.718931 |

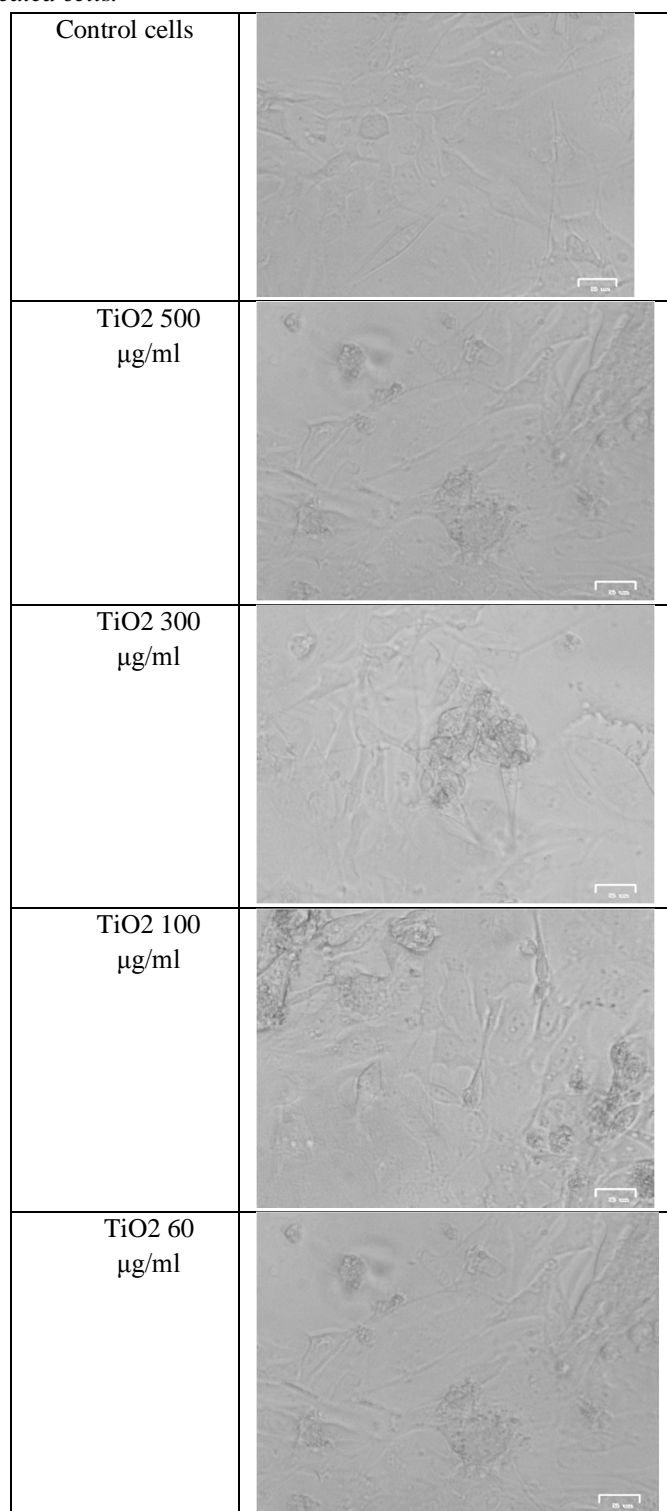
| | | | | | |
|----|-----------|---------|---------|---------|-----------|
| 5 | 200 µg/ml | 50.3479 | 50.3715 | 50.4701 | 50.396483 |
| 6 | 100 µg/ml | 51.839 | 52.0555 | 53.4389 | 52.444444 |
| 7 | 80 µg/ml | 54.2247 | 54.1852 | 54.4285 | 54.279464 |
| 8 | 60 µg/ml | 56.3121 | 57.999 | 58.0406 | 57.45057 |
| 9 | 40 µg/ml | 58.4493 | 60.2278 | 61.4547 | 60.043955 |
| 10 | 20 µg/ml | 61.829 | 63.8435 | 63.7803 | 63.15094 |
| 11 | 10 µg/ml | 70.6262 | 71.372 | 73.33 | 71.776081 |

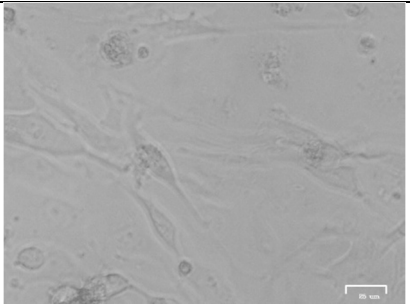
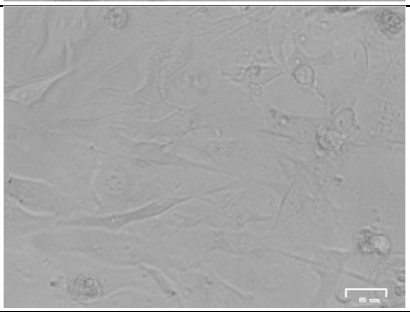


3) IC50 Value of tested sample is 44.85 µg/ml

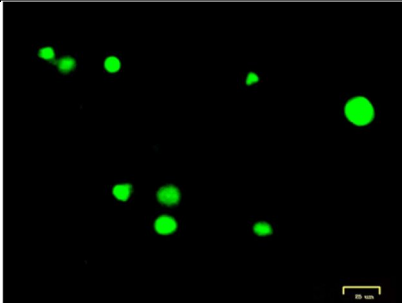
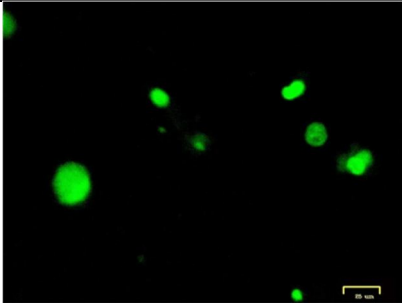
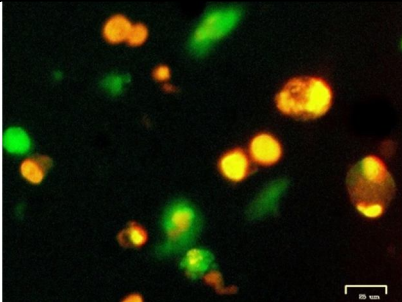
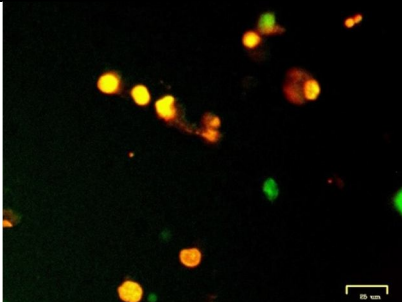
| log(inhibitor) vs. normalized response -- Variable slope | |
|--|-------------------|
| Best-fit values | |
| LogIC50 | 1.652 |
| HillSlope | -1.107 |
| IC50 | 44.85 |
| Std. Error | |
| LogIC50 | 0.03408 |
| HillSlope | 0.1028 |
| 95% CI (asymptotic) | |
| LogIC50 | 1.582 to 1.722 |
| HillSlope | -1.318 to -0.8966 |
| IC50 | 38.19 to 52.67 |
| Goodness of Fit | |
| Degrees of Freedom | 28 |
| R squared | 0.9189 |
| Sum of Squares | 2025 |
| Sy.x | 8.505 |
| Number of points | |
| # of X values | 30 |
| # Y values analysed | 30 |

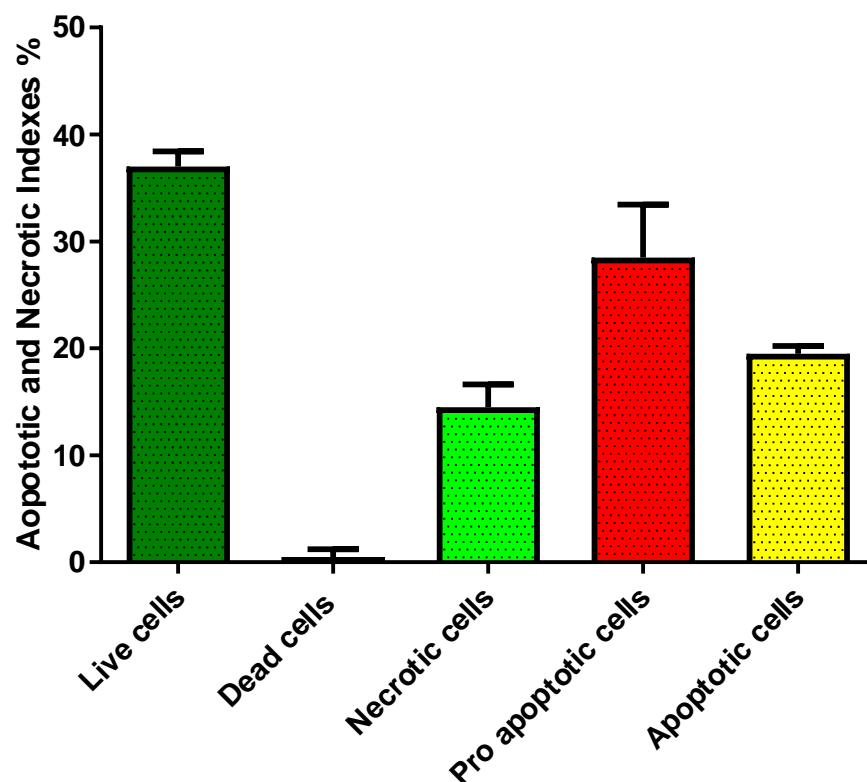
4) *Images of control cells and treated cells.*



| | |
|------------------------------|--|
| TiO ₂ 20 μg/ml |  |
| TiO ₂ 10 μg/ml |  |

C. ETBR/AO STAINING:

| Samples | Fluorescent images | Fluorescent images |
|---|---|--|
| Control |  |  |
| Treated with 44.85 μg/ml of TiO ₂ |  |  |



| S.No | Dead cells | Necrotic cells | Pro-Apoptotic cells | Apoptotic cells | Live cells |
|------|------------|----------------|---------------------|-----------------|------------|
| 1. | 1 | 16 | 25 | 20 | 38 |
| 2. | 0 | 13 | 32 | 19 | 36 |

IV. CONCLUSION

The present study discover that the extracts of *Calotropis gigantea* exhibited anticancer activity on HepG2 cellular lines. Thus the result further studies are needed to investigate and isolate its energetic compounds and also to assess its anticancer potential on other cellular lines.

V. ACKNOWLEDGEMENT

We acknowledge TRI-BIOTECH (Trichy Research Center) for characterization studies.

VI. CONFLICT OF INTEREST

The authors declare that they have“ No Conflict of Interest”.

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