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Formulation and Development of Solid Nano-Emulsifying Drug Delivery System of BCS Class IV Drug Delafloxacin

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Abstract: Aqueous solubility of drugs is one of the most important factors that determine their dissolution performance and hence oral absorption and bioavailability. About 70% of new drug substances are poorly water soluble and exhibit slow dissolution rates and often incomplete oral bioavailability. The aim of present investigation is to develop a novel emulsifying drug delivery system (SNEDDS) to enhance the dissolution rate profile by increasing the solubility of Delafloxacin. The oral route is the most commonly used method for administration of drugs, with nearly 80% of the marketed dosage forms being delivered orally. This route of drug administration is the most convenient and non-invasive, leading to better patient compliance. Therefore, different formulation strategies have been investigated for enhancing the solubility of poorly soluble drugs with the aim of improving oral bioavailability. However, the optimized formulation has chosen a suitable emulsifying rate with a quick dissolution rate based on the smaller z-average diameter. The optimized Delafloxacin SNEDDS has exhibited faster dissolution rate as 97 ± 2.07 % drug released which is clearly indicating that developed SNEDDS enhanced the dissolution rate of the drug.

Keywords: SNEDDS, Emulsifying, Delafloxacin, Bioavailability, Dissolution, Enhancing.

I. INTRODUCTION

Delafloxacin is a fourth generation fluoroquinolone antibiotic which has been used in trials studying the treatment and basic science of Gonorrhoea, Hepatic Impairment, Bacterial Skin Diseases, Skin Structure Infections, and Community Acquired Pneumonia, among others.^{1,2} Comparing to other FQs, a major structural substitute change on C-7 makes DLF more powerful against gram positive and gram negative bacteria. Due to the lack hydrogen ion substitute in the chemical structure of DLF, it got the distinct property of weakly acidic moiety with enhanced intracellular permeation and potent bactericidal effects. The maximum plasma concentration (C_{max}) of DLF was achieved between 1 and 2.5 h in healthy volunteers and the reported absolute bioavailability of DLF is 58.8% only which might be due to its poor solubility profile. Poorly water-soluble drugs frequently display problems of low solubility and slow dissolution rate in gastrointestinal (GI) media which results in low and variable oral absorption. Therefore, different formulation strategies have been investigated for enhancing the solubility of poorly soluble drugs with the aim of improving oral bioavailability.^{3,4}

II. MATERIALS AND METHODS

A. Materials

Delafloxacin was obtained as gift sample from Vama Pharma (Nagpur). Capryol 90, Poloxamers 188 Span 20, Tween 20 etc was obtained from Anand agencies Pune. All chemicals used were analytical grade.

B. Methods

1) Preformulation Study

a) Melting Point

The melting point determined by Thieles tube method. The results were reported in results part.

b) Solubility study

Solubility study of Delafloxacin were carried out in different volatile solvent i.e., water, 0.1 N HCl and 6.8 pH phosphate buffer. Saturated solutions were prepared by adding drug in excess amount to the vehicle for 48 hr at 25 °C under constant stirring.

After this period the solution were filtered, diluted with distilled water (at least 1000 times) and analyzed by UV spectrophotometer at a wavelength of 298 nm.

2) *UV Spectrophotometric Study*

a) *Preparation of calibration curve of Delafloxacin*

A stock standard solution (1 mg/ml) of Delafloxacin was prepared in Potassium Phosphate buffer 3.6 and diluted with ion-free water to prepare working solutions at concentrations of 0, 10, 20, 30, 40, 50 µg/ml respectively. The absorbance of standard solution was determined using UV/ VIS spectrophotometer at 298 nm.

b) *Study of Delafloxacin Solubility in Various Oils, Surfactants and Co-surfactants*

In order to find out the right SNEDDS components with good solubilizing capacity for Delafloxacin, saturation solubility was performed on different oils (Capryol 90, Algae oil, Palm oil, and peanut oil), surfactants (Poloxamer 188, Cremophor S9, Labrasol, Tween 20, Span 20) and co-surfactants (Transcutol HP, PEG 400, Propylene glycol and Glycerin) using the shake flask method. All measurements were done in triplicate and the solubility was expressed as the mean value (mg/ml) ±SD.

3) *Preparation of Delafloxacin loaded SNEDDS*

a) Nine different formulation of Delafloxacin loaded SNEDDS were prepared by composition shown in table.

b) Accurately weighed of surfactant (Tween-20, span-20, Poloxamer 188) and co-surfactant (Transcutol HP) were mixed in a vial on magnetic stirrer. The weighed amount of drug Delafloxacin was dissolved in the selected amount of oil (Capryol 90 oil) in a separate beaker. The oil phase was added drop wise to the surfactant co-surfactant mix and stirring was continued for 1 h to obtain Delafloxacin loaded liquid SNEDDS.

c) S-SNEDDS of Delafloxacin was prepared by adsorption method. Optimized L-SNEDDS preparation was transformed to S-SNEDDS by the adsorption method. The benefit of the adsorption technique is to have more surface area with good physical adsorption, high dissolution efficiency, good uniformity, and reproducibility. In a glass mortar, Delafloxacin loaded L-SNEDDS was added drop wise in 1:1 w/w proportion with adsorbent Aerosil. After every addition, the blend was mixed properly until a uniform solid powder was obtained. The resultant damp mass was collected from the apparatus and processed through the mesh size 250 microns (mesh no 60). solid SNEDDS was placed at 25°C in desiccators for further characterizations.

Table No. 1: Formulation of Delafloxacin SNEEDS

| | F1 | F2 | F3 | F4 | F5 | F6 | F7 | F8 | F9 |
|---------------------------|-----|-----|-----|-----|-----|-----|-----|-----|-----|
| Delafloxacin (mg) | 800 | 800 | 800 | 800 | 800 | 800 | 800 | 800 | 800 |
| Capryol 90 (ml) | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 | 1 |
| Tween 20 (mg) | 100 | 200 | 300 | - | - | - | - | - | - |
| Poloxamers 188 (mg) | - | - | - | 100 | 200 | 300 | - | - | - |
| Span 20 (mg) | - | - | - | - | - | - | 100 | 200 | 300 |
| Transcutol HP (mg) | 50 | 100 | 150 | 50 | 100 | 150 | 50 | 100 | 150 |
| Ethanol (Co-solvent) (ml) | 25 | 25 | 25 | 25 | 25 | 25 | 25 | 25 | 25 |

III. EVALUATION OF DELAFLOXACIN LOADED S-SNEDDS

1) Percent transmittance

The percent transmittance was measured at 298 nm using UV-Vis spectrophotometer against double distilled water as the blank. The studies were conducted at 50, 100, 250 and 1000 times dilution.

2) Zeta potential

The zeta potential of the diluted SNEDDS formulations was measured using a Malvern zetasizer. The SNEDDS were diluted with a ratio of 1:2500 (v/v) with distilled water and mixed for 1 min using a magnetic stirrer. Zeta potential of each SNEDDS was determined in triplicate.

3) Mean globule size

The formulation, 100 μ L was diluted to 250 mL in a beaker and gently mixed using a glass rod. The resultant emulsion was then subjected to particle size analysis using Malvern zetasizer. All studies were repeated for six times, with good agreement being found between measurements.

4) Polydispersity Index

The efficiency and dispersibility of self-emulsification were determined through USP dissolution apparatus 2. Briefly, 1 mL of each formulation was added drop wise into 200 mL of simulated intestinal fluid (pH 6.8, without enzymes), maintained at 37 °C, with gentle stirring using stainless steel paddles rotated at 60 rpm.

5) Viscosity

The viscosity of Delafloxacin SNEDDS was determined with Brookfield viscometer (digital viscometer + Pro) at 20 rpm at room temperature (25°C)

6) Cloud point measurement

The SNEDDS formulations were compared for cloud point value. Each formulation was diluted with water in the ratio of 1:100 and placed in a water bath with gradual increase in temperature. At the cloud point, drop in the percent transmittance of sample from the zero point was measured spectrophotometrically.

7) Micromeritic Properties of S-SNEDDS

The values obtained for the angle of repose of the S-SNEDDS formula F5 was $27.64^{\circ} \pm 1.03^{\circ}$, as shown in Table. This value indicate that formula have good Flowability. The bulk density of the formula F5 was found to be 0.48 ± 0.03 g/ml. However, tapped density was 0.57 ± 0.021 g/mL for formula F5. Carr's index of formula F5 was found to be 14.22 ± 1.62 which give an indication about the good Flowability of the S-SNEDDS formula. This was further supported by the values of Hausner's ratio. The results of Hausner's ratio of formula F5 was 1.24 ± 0.10 . The improved Flowability of S-SNEDDS formula may be due to good sphericity of particles.

8) Drug Loading Efficiency/Drug content

Samples were prepared in triplicate and absorbance was measured after suitable dilutions at 298 nm using UV-Vis Spectrophotometer. The amount of Delafloxacin present in each formula was calculated from a calibration plot.

9) Scanning Electron Microscopy (SEM)

Scanning electron micrographs for Delafloxacin, prepared S-SNEDDS formulae were taken using Scanning electron microscope operating at 20 kV to study surface topography of S-SNEDDS. The samples were fixed on SEM stub and then coated with thin layer of platinum.

10) Differential Scanning Calorimetry (DSC)

Thermograms of Delafloxacin optimized S-SNEDDS formulae were obtained using differential scanning calorimeter. The thermal behavior was studied by heating nearly 2 mg of samples in sealed aluminum pans under nitrogen gas flow (30 ml/min) over a temperature range of 0 to 250°C and a heating rate of 10°C/min.

11) Fourier Transformed Infrared Spectroscopy (FTIR)

FTIR Spectra of pure Delafloxacin prepared and optimized S-SNEDDS formulae were obtained using Fourier transformed infrared spectrophotometer. Solid samples were mixed with small quantity of IR grade potassium bromide and compressed into discs by applying pressure. The compressed disc was placed in light path and the spectrum was obtained. Each KBr disc was scanned at 4 mm/s at a resolution of 2 cm over a wave number region of 4000-400 cm^{-1} .

12) In Vitro Drug Release Studies

The in vitro drug release of Delafloxacin from the optimized SNEDDS formulation, pure drug and marketed product was performed utilizing USP disintegration apparatus type II. The dissolution medium comprised of 900 mL of newly pre-arranged phosphate buffer pH 3.6 kept up with at $37 \pm 0.5^{\circ}\text{C}$ and the paddle speed was set at 50 rpm.

Hard gelatin cases, size "000" stacked up with pre concentrate were joined to paddles using Para film spring to hold cases back from floating. Aliquots (5 mL) from the disintegration medium were removed at standard time stretches (5, 10, 15, 30, 45, 60, 90 and 120 min) utilizing an adjusted expendable needle. The examples were then separated through a film channel (0.45 μm , Whatman) and medication fixation was gotten after appropriate weakening by means of UV approved technique at 298 nm using UV-Vis Spectrophotometer.

IV. STABILITY STUDIES

The SNEDDS formulations were filled into empty hard gelatin capsules (size 0) and subjected to stability studies at 25°C/60 % relative humidity (RH) and 40°C/75% RH. Samples were charged in stability chambers with humidity and temperature control. They were withdrawn at specified intervals for analysis over a period of 3 month.

V. RESULTS AND DISCUSSIONS

1) Preformulation Study

Table No. 2: Organoleptic Properties of Delafloxacin

| Test | Specification | Observation |
|-------|---------------------------|---------------------------|
| Color | White to pinkish crystals | White to pinkish crystals |
| Odor | Odorless | Odorless |

2) Melting Point

Table No. 3: Melting Point of Delafloxacin

| Drug Name | Standard Melting Point | Observation Melting Point |
|--------------|------------------------|---------------------------|
| Delafloxacin | 224-227 °C | 227 °C |

3) Solubility

The solubility of Delafloxacin in different solvent Water, 0.1 N HCl and Phosphate buffer pH 3.6 was determined.

Table No. 4: Solubility of Delafloxacin in Different solvent

| Medium | Delafloxacin (mg/ml) |
|-------------------------|----------------------|
| Water | 0.122 |
| 0.1 N HCl | 8.72 |
| Phosphate buffer pH 3.6 | 4.31 |

4) Calibration Curve of Delafloxacin

The maximum of Delafloxacin was found at 298 nm. The linear relationship was observed over the range of 0-50 $\mu\text{g/ml}$. Absorbance were noted at 298 nm against Potassium Phosphate buffer pH 3.6 as a blank. A calibration graph of the absorbance versus drug concentration of the drug was plotted.

Table No. 5: Calibration curve for Delafloxacin

| Sr. No. | Concentration($\mu\text{g/ml}$) | Absorbance (nm) |
|---------|-----------------------------------|-----------------|
| 1 | 0 | 0 |
| 2 | 10 | 0.215 |
| 3 | 20 | 0.414 |
| 4 | 30 | 0.601 |
| 5 | 40 | 0.732 |
| 6 | 50 | 0.907 |
| | R ² | 0.994 |

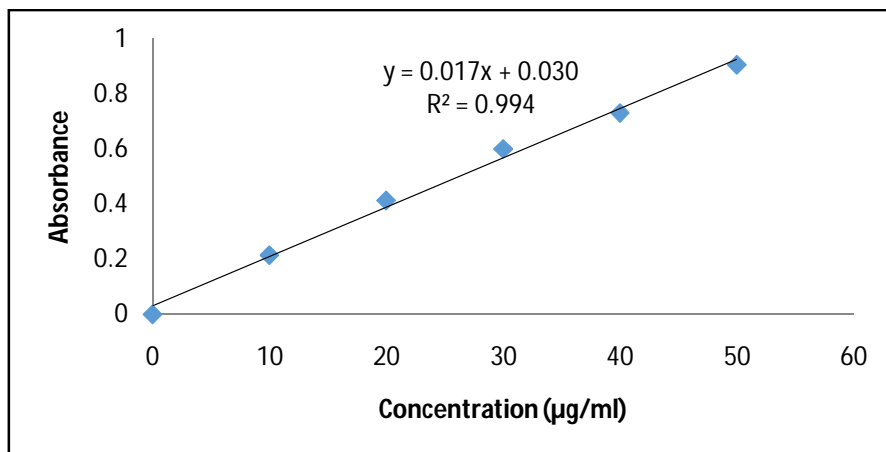


Figure No.1: Calibration curve of Delafloxacin

5) Study of Delafloxacin Solubility in Various Oils, Surfactants and Co-surfactants

The results of solubility studies of drug in various oils, surfactants and co-surfactants are as shown in Table. These highlighted ingredients were selected for further study of emulsification ability and miscibility with other ingredient

Table No 6: Solubility of Delafloxacin Solubility in Various Oils, Surfactants and Co-surfactants

| Category | Name of excipient | Solubility (mg/ml) |
|----------------|-------------------|--------------------|
| Oil | Palm oil | 102.44± 1.43 |
| | Capryol 90 | 136.72± 2.21 |
| | Algae oil | 98.04 ± 2.01 |
| | Peanut oil | 74.36 ± 1.36 |
| Surfactant | Poloxamer 188 | 307.63±3.18 |
| | Cremophor S9 | 69.29±3.45 |
| | Labrasol | 38.57±3.76 |
| | Tween 20 | 304.98±3.03 |
| Co-surfactants | Span 20 | 161.34±1.34 |
| | Transcutol HP | 99.09±3.52 |
| | PEG 400 | 80.72±2.46 |
| | Propylene glycol | 53.85±1.45 |
| | Glycerin | 44.32±1.23 |

6) Evaluation of S-SNEDDS

The transmittance values of all formulation above 90%, confirming the self-nano-emulsification efficiency of the SNEDDS.

Table No 7: Effect of dilution on different SNEDDS formulations using double distilled water as a dilution media

| Dilution | Percentage of transmittance | | | | | | | | |
|-----------|-----------------------------|--------------|---------------|--------------|--------------|--------------|--------------|--------------|--------------|
| | F1 | F2 | F3 | F4 | F5 | F6 | F7 | F8 | F9 |
| 50 times | 79.23±0.0378 | 77.35±0.078 | 75.23±0.054 | 80.23±0.0571 | 82.23±0.0861 | 81.82±0.0417 | 85.23±0.0527 | 61.89±0.0532 | 69.23±0.0683 |
| 100 times | 84.89±0.0235 | 85.56±0.0124 | 80.89±0.00137 | 85.89±0.0582 | 87.89±0.0196 | 84.23±0.0937 | 87.89±0.0462 | 75.43±0.0237 | 74.89±0.0163 |

| | | | | | | | | | |
|------------|--------------|--------------|--------------|--------------|--------------|--------------|--------------|--------------|--------------|
| 250 times | 88.46±0.0904 | 89.53±0.0457 | 84.46±0.0739 | 88.46±0.0329 | 94.46±0.0644 | 86.46±0.0523 | 91.46±0.0682 | 89.46±0.0567 | 89.46±0.0628 |
| 1000 times | 90.51±0.0670 | 91.81±0.0781 | 90.99±0.0235 | 93.51±0.0523 | 99.51±0.0824 | 94.51±0.0682 | 95.51±0.0435 | 90.34±0.0473 | 94.51±0.0241 |

7) Measurement of Physical Properties of SNEDDS

Physical properties including Z-average diameter, PDI, Zeta potential, viscosity and cloud point of formulation batches F1-F9 were measured.

Table No 8: Interpretation of results of physical properties

| Test parameter | SNEDDS formulation | | | | | | | | |
|---------------------|--------------------|--------------|--------------|--------------|--------------|--------------|--------------|--------------|--------------|
| | F1 | F2 | F3 | F4 | F5 | F6 | F7 | F8 | F9 |
| Z- average diameter | 148.29±0.198 | 182.64±0.157 | 168.58±0.147 | 174.23±0.231 | 124.68±0.145 | 156.75±0.142 | 142.11±0.561 | 161.54±0.142 | 146.51±0.157 |
| PDI | 0.195±0.0119 | 0.173±0.0124 | 0.254±0.201 | 0.169±0.0214 | 0.125±0.0137 | 0.578±0.0657 | 0.565±0.0632 | 0.531±0.0271 | 0.651±0.0652 |
| Zeta potential (mV) | -4.84 | -2.53 | -4.18 | -3.75 | -5.74 | -6.57 | -4.63 | -4.35 | -3.46 |
| Viscosity (cPs) | 36 | 59 | 39 | 48 | 24 | 34 | 45 | 54 | 38 |
| Cloud point (°C) | 76-77 | 75-77 | 75-76 | 74-75 | 77-78 | 73-74 | 73-75 | 75-76 | 75-77 |

8) Drug Loading Efficiency/Drug content

The amount of Delafloxacin present in the optimized SNEDDS formulae was found to be within the USP limit. The drug content in SNEDDS was almost identical with the results obtained in SNEDDS, so there was no change of percentage drug content.

Table No. 9: % Drug loading efficiency

| Formula | Drug loading efficiency |
|---------|-------------------------|
| F1 | 92.37±0.75 |
| F2 | 93.30 ±0.86 |
| F3 | 96.85± 1.23 |
| F4 | 95.22 ±2.22 |
| F5 | 99.09± 0.56 |
| F6 | 96.32±1.88 |
| F7 | 98.53± 1.45 |
| F8 | 94.68±1.91 |
| F9 | 95.76±1.64 |

9) Scanning Electron Microscopy (SEM)

The surface morphology of pure Delafloxacin powder and S-SNEDDS formulae of Delafloxacin was determined using scanning electron microscope. The Delafloxacin powder appeared with an irregular crystalline shape as irregular and plate-shaped crystals having rough surfaces.

The image of the solid SNEDDS formulae F-5 containing Delafloxacin however, illustrate that the particles had the same outer macroscopic morphology consisting of well separated spherical particles with relatively deep dents and similar diameters.

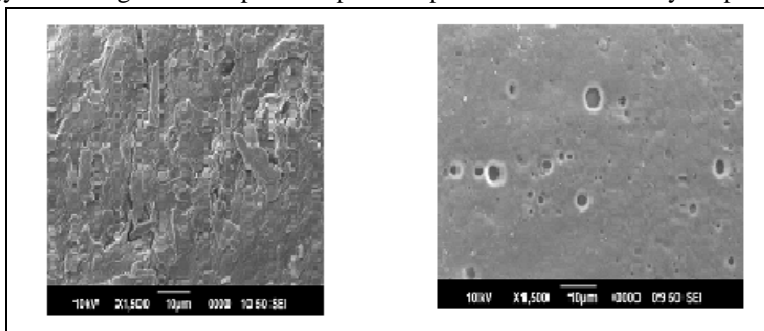


Figure No. 2: Surface morphology of pure Delafloxacin powder

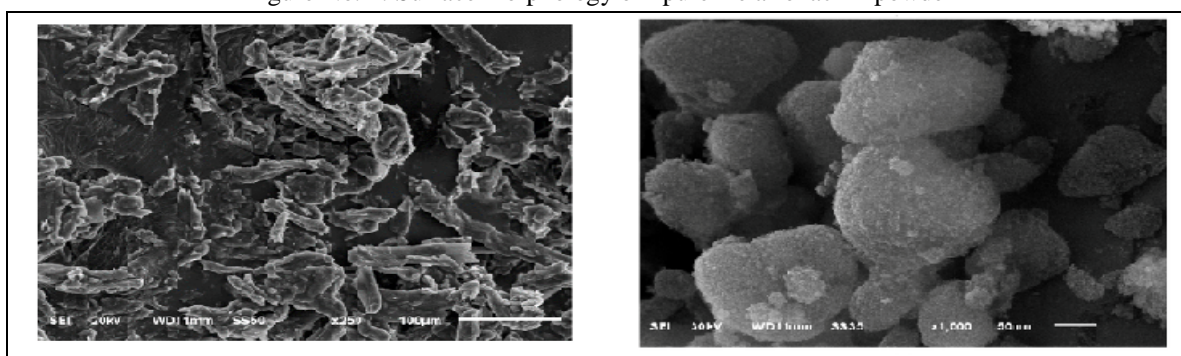


Figure No. 3: Surface morphology of S-SNEDDS formulae

10) Differential Scanning Calorimetry (DSC)

Thermograms of pure Delafloxacin, and prepared optimized S-SNEDDS F-5 were obtained using differential scanning calorimeter. It is known that transforming the physical state of a drug to the amorphous or partially amorphous state leads to a high-energy state and high disorder, resulting in enhanced solubility. As a result, it was expected that the solid particles would also have enhanced solubility.

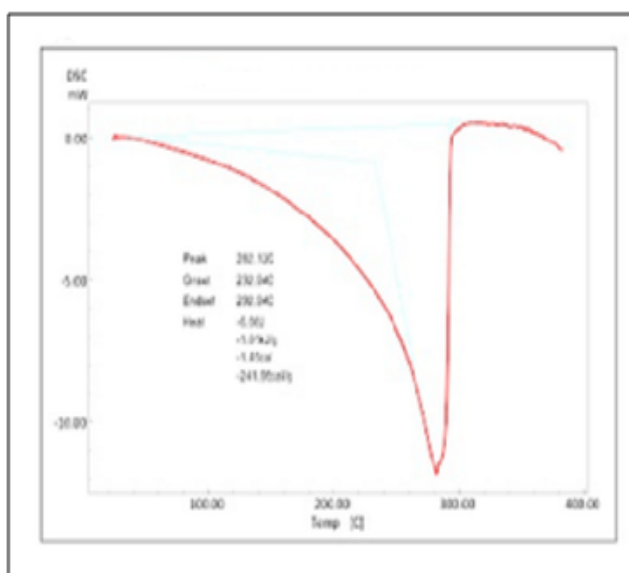


Figure No 4: Thermograms of Pure Delafloxacin

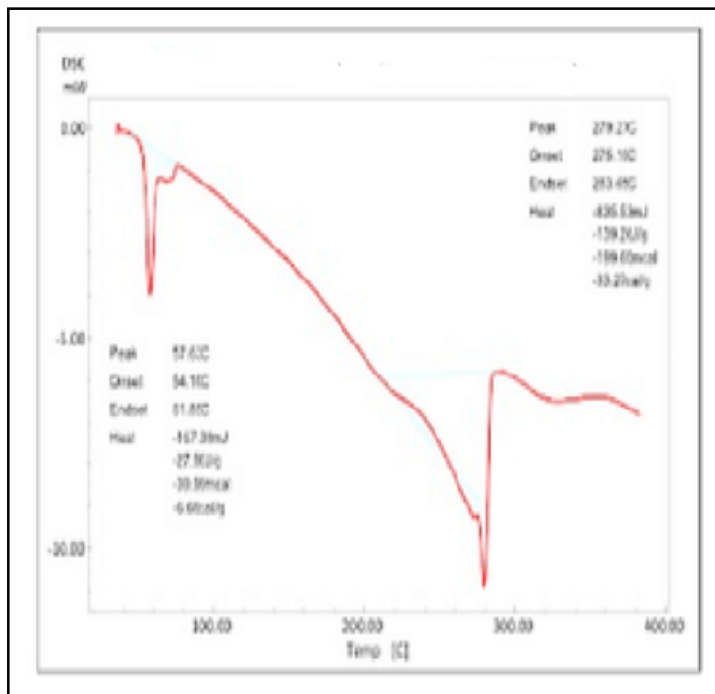


Figure No 5: Thermograms of optimized S-SNEDDS F-5

11) Fourier Transformed Infrared Spectroscopy (FTIR)

FTIR spectra are mainly used to determine interaction between the drug and any of the excipients used. The presence of interaction is detected by the disappearance of the important functional group of the drug.

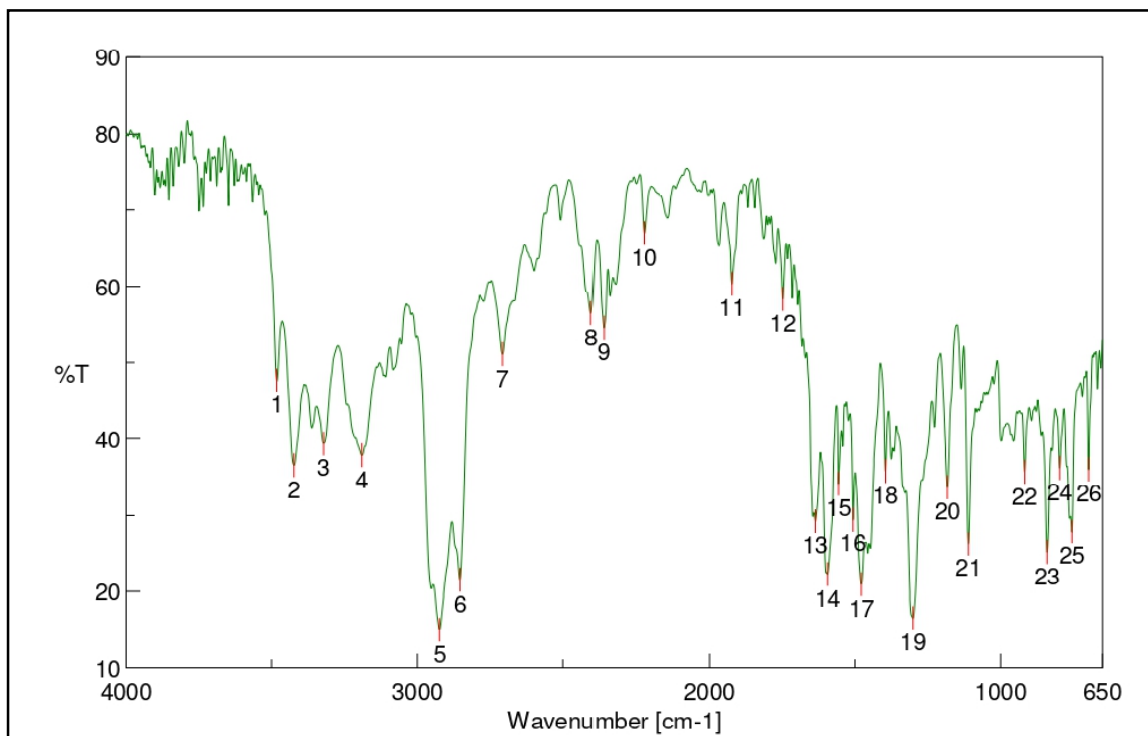


Figure No 6: FTIR spectra of pure drug

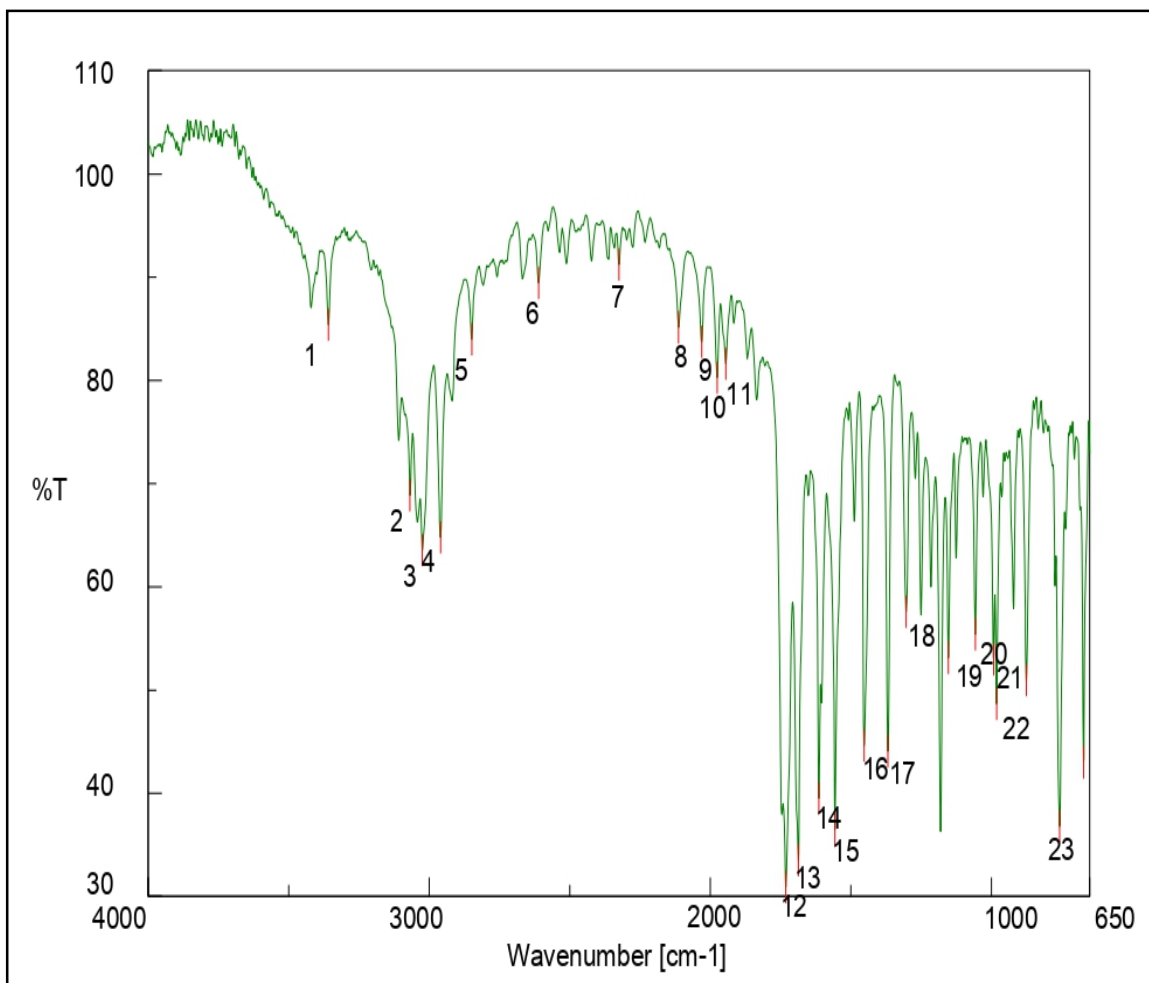


Figure No 7: FTIR spectra of optimized S-SNEDDS F-5

12) *In Vitro* Drug Release Studies

The percentage drug release from S-SNEDDS was found to be higher than that of pure Delafloxacin and marketed product as shown.

Table No. 10: In Vitro Drug Release

| Time (h) | Cumulative% drug release | | | | | | | | |
|----------|--------------------------|----------|---------|----------|----------|---------|---------|---------|---------|
| | F1 | F2 | F3 | F4 | F5 | F6 | F7 | F8 | F9 |
| 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 | 0 |
| 5 | 11 ±1.07 | 12 ±2.47 | 20±2.47 | 10 ±1.17 | 10 ±2.97 | 14±1.63 | 20±2.54 | 18±2.35 | 18±1.55 |
| 10 | 22±3.56 | 21±3.25 | 28±2.94 | 23±1.34 | 21±2.29 | 26±3.25 | 32±2.66 | 27±2.57 | 33±2.88 |
| 15 | 31±2.78 | 32±3.59 | 33±3.78 | 32±2.52 | 33±3.33 | 39±3.68 | 43±2.70 | 36±3.61 | 44±2.63 |
| 30 | 42±3.38 | 44±2.80 | 44±4.58 | 41±3.16 | 43±2.71 | 50±3.67 | 55±2.37 | 47±3.95 | 53±3.85 |
| 45 | 54±4.14 | 56±3.11 | 53±3.84 | 55±3.74 | 53±4.04 | 63±2.81 | 70±3.91 | 55±3.68 | 64±4.08 |
| 60 | 65±3.90 | 68±4.08 | 62±4.52 | 64±3.81 | 64±3.81 | 78±3.58 | 78±2.46 | 67±2.74 | 75±3.73 |
| 90 | 79±3.14 | 70±2.04 | 80±2.28 | 74±2.24 | 71±3.50 | 90±1.74 | 87±2.77 | 83±2.37 | 82±1.88 |
| 120 | 90±1.78 | 93±2.16 | 86±2.36 | 94±1.17 | 97±2.07 | 92±1.93 | 95±1.30 | 92±1.91 | 90±2.51 |

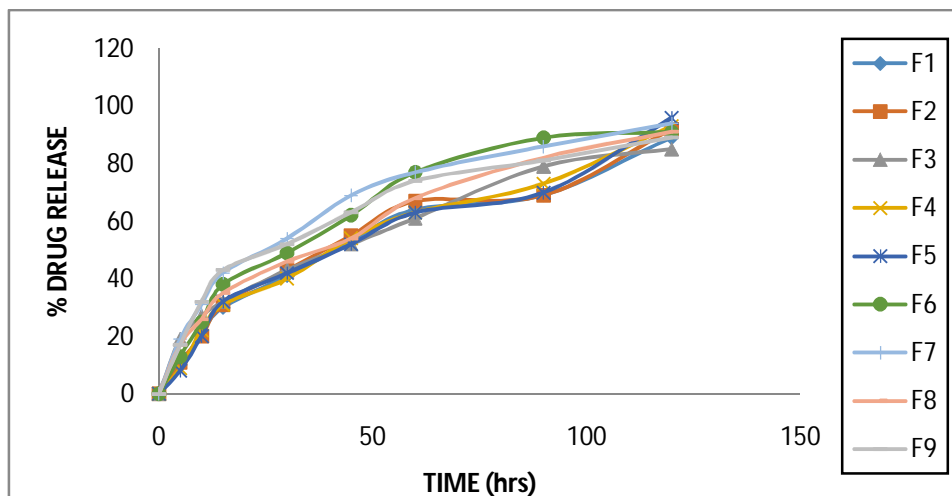


Figure No. 8: Cumulative % drug release

Delafloxacin dissolved and released from S-SNEDDS reached 97.02 % \pm 2.07 % for formula F-5, within two hour.

13) Mathematical Model Approach

Dissolution data of the formulations were fitted to various mathematical models (zero-order, first-order, Peppas, and Hixon–Crowel) in order to describe the kinetics of drug release. The smallest value of the sum of squared residuals (SSR) and the highest value of the correlation coefficient (r) were taken as criteria for selecting the most appropriate model. It is clear that optimized formulas have R^2 value 0.977, indicating Fickian diffusion-controlled release

Table No 11: In vitro drug release studies (F1-F9)

| Formulation Code | KINETIC MODELS | | | | |
|------------------|------------------|-------------------|---------------|-----------------|-----------------------------------|
| | Zero order R^2 | First order R^2 | Higuchi R^2 | Korsmeyer R^2 | Best Fitted Model (Higuchi) R^2 |
| F1 | 0.918 | 0.956 | 0.986 | 0.934 | 0.986 |
| F2 | 0.910 | 0.928 | 0.980 | 0.932 | 0.980 |
| F3 | 0.913 | 0.989 | 0.996 | 0.868 | 0.996 |
| F4 | 0.934 | 0.937 | 0.987 | 0.939 | 0.987 |
| F5 | 0.932 | 0.865 | 0.977 | 0.944 | 0.977 |
| F6 | 0.867 | 0.979 | 0.977 | 0.912 | 0.977 |
| F7 | 0.843 | 0.992 | 0.980 | 0.865 | 0.980 |
| F8 | 0.912 | 0.991 | 0.996 | 0.887 | 0.996 |
| F9 | 0.830 | 0.980 | 0.974 | 0.861 | 0.974 |

14) Stability Studies

Table No. 12: Stability Study of Delafloxacin S-SNEDDS

| Days | Temp. condition | PDI | % of transmittance | Drug content |
|------|-----------------|-----------------|--------------------|--------------|
| 0 | 2-8°C | 0.125±0.01370.1 | 99.51±0.84 | 99.09± 0.56 |
| | 30±2°C/65±5% RH | 25±0.0137 | 99.51±0084 | 99.09± 0.56 |
| | 40±2°C/75±5% RH | 0.125±0.0137 | 99.51±0.84 | 99.09± 0.56 |
| 30 | 2-8°C | 0.142±0.01760.1 | 99.79±0.15 | 99.55± 0.015 |
| | 30±2°C/65±5% RH | 67±0.0136 | 99.85±0.81 | 99.29± 0.008 |
| | 40±2°C/75±5% RH | 0.179±0.0114 | 99.46±0.25 | 99.15± 0.011 |
| 60 | 2-8°C | 0.150±0.01870.1 | 99.83±0.18 | 99.39± 0.010 |
| | 30±2°C/65±5% RH | 87±0.0127 | 99.89±0.09 | 99.19± 0.18 |
| | 40±2°C/75±5% RH | 0.195±0.0125 | 99.51±0.13 | 99.93± 0.20 |
| 90 | 2-8°C | 0.125±0.01370.1 | 99.80±0.015 | 99.32± 0.013 |
| | 30±2°C/65±5% RH | 25±0.0137 | 99.85±0.012 | 99.89± 0.014 |
| | 40±2°C/75±5% RH | 0.125±0.0137 | 99.69±0.012 | 99.79± 0.006 |

Physical appearance of SNEDDS was observed during the stability studies and the SNEDDS remained clear with no signs of precipitation. This indicated that the drug remained solubilized even at accelerated stability conditions (40 ± 2°C/75 ± 5% RH). Also, no significant decrease in the Delafloxacin content was observed indicating that drug remained chemically stable in the SNEDDS. Thus, it can be concluded that the Delafloxacin SNEDDS would remain physicochemical stable at long-term stability conditions (30 ± 2°C/65 ± 5% RH) as well as accelerated conditions (40 ± 2°C/75 ± 5% RH) for 3 months.

VI. CONCLUSION

Delafloxacin is a lipophilic drug as constrained of aqueous solubility, hepatic first-pass metabolism with poor absorption rate thus, formulated to overcome such hindrances. The developed liquid formulations have found appropriate results of thermodynamic stability, emulsifying rate and adequate dissolution rate. Whereas, based on the characterization and evaluation of numerous parameters have revealed that the developed formulations were providing an efficacious therapeutic efficacy of the drug. The optimized Delafloxacin SNEDDS has selected based on the smaller z-average diameter and faster dissolution rate as well as an appropriate emulsification rate as compared with the other formulations. The optimized formulation had exhibited faster dissolution rate as 97±2.07 % drug released which is clearly indicating that developed SNEDDS enhanced the dissolution rate of the drug. Therefore, It reveals that the characterization, evaluation and comparison of various parameters have to reached on the conclusion that S-SNEDDS is a potential drug delivery system of the drug. Finally, we can conclude that S-SNEDDS is a robust and promising nanocarriers to enhance drug bioavailability and therapeutic efficacy.

REFERENCES

- [1] Shah, M.K.; Khatri, P.; Vora, N.; Patel, N.K.; Jain, S.; Lin, S. Lipid Nanocarriers: Preparation, Characterization and Absorption Mechanism and Applications to Improve Oral Bioavailability of Poorly Water-Soluble Drugs. In Biomedical Applications of Nanoparticles; William Andrew Publisher: Norwich, NY, USA, 2019; pp. 117–147, ISBN 9780128165065.
- [2] Boyd, B.J.; Bergström, C.A.S.; Vinarov, Z.; Kuentz, M.; Brouwers, J.; Augustijns, P.; Brandl, M.; Bernkop-Schnürch, A.; Shrestha, N.; Pr at, V.; et al. Successful Oral Delivery of Poorly Water-Soluble Drugs Both Depends on the Intraluminal Behavior of Drugs and of Appropriate Advanced Drug Delivery Systems. Eur. J. Pharm. Sci. 2019, 137, 104967
- [3] Kalepu, S., Manthina, M. &Padavala, V. 2013. Oral lipid-based drug delivery systems – an overview. ActaPharmaceuticaSinica B, 3 (6), pp. 361-372.
- [4] Krishnaiah, Y. S. R. 2010. Pharmaceutical Technologies for Enhancing Oral Bioavailability of Poorly Soluble Drugs. Journal of Bioequivalence & Bioavailability, 2 (2), pp. 028-036.
- [5] Rajendra C., Harish P. Jain A.K., Himesh S., SaraogiG.K.(2011). Preparation and evaluation of the self-emulsifying drug delivery system containing atorvastatin HMGCOA inhibitor. International Journal of Pharmacy and Pharmaceutical Sciences, 3 (3): 147-152., ISSN- 0975-1491.
- [6] Lipinski, C. A. (2000).Drug-like properties and the causes of poor solubility and poor permeability. J PharmacolToxicol Methods, 44 (1), 235-249., ISSN: 1056-8719.
- [7] Ghose, A. K., Viswanadhan, V. N., &Wendoloski, J. J. (1999).A Knowledge-Based Approach in Designing Combinatorial or Medicinal Chemistry Libraries for Drug Discovery. 1. A Qualitative and Quantitative Characterization of Known Drug Databases. Journal of Combinatorial Chemistry, 1 (1), 55-68., ISSN: 2156-8952.



- [8] Shahba, A. A., Mohsin, K., & Alanazi, F. K. (2012). Novel Self-Nanoemulsifying Drug Delivery Systems (SNEDDS) for Oral Delivery of Cinnarizine: Design, Optimization, and In-Vitro Assessment. *AAPS PharmSciTech*, 13 (3), 967-977., ISSN: 1530-9932.
- [9] Sven S., Hassan B., Thilo S., Martin O. (2009). Self-Emulsifying Drug Delivery Systems Facing the bioavailability challenge in drug delivery. *Die Pharmazeutische Industrie*, 71(8):1409-1416., ISSN 0031-711X.
- [10] Gursoy, R. N., & Benita, S. (2004). Self-emulsifying drug delivery systems (SEDDS) for improved oral delivery of lipophilic drugs. *Biomed Pharmacother*, 58 (3), 173-182. ISSN: 0753-3322.
- [11] Stegemann, S., Leveiller, F., Franchi, D., de Jong, H., & Linden, H. (2007). When poor solubility becomes an issue: from early stage to proof of concept. *Eur J Pharm Sci*, 31(5), 249-261., ISSN: 0928-0987.
- [12] Mangale M. R., Pathak S, Mene H. R, More B. (2015). A nanoemulsion, as pharmaceutical overview. *Int. J. Pharm. SCI. Rev. Res.*, 33 (1), 244-252., ISSN 0976 – 044X
- [13] Mahmoud, D. B., Shukr, M. H., & Bendas, E. R. (2014). In vitro and in vivo evaluation of self-nanoemulsifying drug delivery systems of cilostazol for oral and parenteral administration. *Int J Pharm*, 476(1-2), 60-69.
- [14] Jiajia R., David J. M.(2012). Lemon oil solubilization in mixed surfactant solutions: Rationalizing microemulsion and nanoemulsion formation. *Food Hydrocolloids*. 26(1), 268-276.
- [15] Avachat, A. M., & Patel, V. G. (2015). Self nanoemulsifying drug delivery system of stabilized ellagic acid-phospholipid complex with improved dissolution and permeability. *Saudi Pharm J*, 23(3), 276-289.
- [16] Dunnhaupt, S., Kammona, O., Waldner, C., Kiparissides, C., & Bernkop-Schnurch, A. (2015). Nano-carrier systems: Strategies to overcome the mucus gel barrier. *Eur J Pharm Biopharm*, 96, 447-453.
- [17] Karavasili C, Andreadis I, Tsantarliotou M, Taitzoglou I, Chatzopoulou P, Katsantonis D, Zacharis C, Markopoulou C, Fatouros D. Self-Nanoemulsifying Drug Delivery Systems (SNEDDS) Containing Rice Bran Oil for Enhanced Fenofibrate Oral Delivery: In Vitro Digestion, Ex Vivo Permeability, and In Vivo Bioavailability Studies. *AAPS PharmSciTech*. 2020; 21(6).
- [18] Beg S, Swain S, Singh HP, PatraChN, Rao ME. Development, optimization, and characterization of solid self-nanoemulsifying drug delivery systems of valsartan using porous carriers. *AAPS PharmSciTech*. 2012 Dec;13(4):1416-27.
- [19] Beg S, Sandhu PS, Batra RS, KhuranaRK, Singh B. QbD-based systematic development of novel optimized solid self-nanoemulsifying drug delivery systems (SNEDDS) of lovastatin with enhanced biopharmaceutical performance, *Drug Delivery*, 2015; 22(6):765-784.
- [20] Fayed ND, GodaAE, Essa EA, Maghraby MEI. Chitosan-encapsulated niosomes for enhanced oral delivery of atorvastatin. 2021; 66: 102866.
- [21] Verma R, Kaushik D, Kumar M, Parmar D. Components of self-microemulsifying drug delivery systems. *A Comprehensive Textbook on self-emulsifying Drug Delivery Systems*. 2021: 79-94



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