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## International Journal for Research in Applied Science & Engineering Technology (IJRASET) Enhancement of Heat Transfer in Shell and Tube Heat Exchanger Using MGO Nano Fluid

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Abstract- Heat exchangers are most widely used devices for industrial purposes. Because there is a lot of need for cooling as well as to dissipate heat. So, here in this paper we aim at rapid cooling and high heat dissipation. For cooling purpose, we generally use heat exchangers with cooling medium as water, ethylene glycol, etc. Now to enhance heat transfer rate in linear flow we are adding Nano particles to the base fluid (water) such that it becomes nanofluid and trying to obtain maximum heat transfer rate in shell and tube heat exchanger. After all practicals done we obtain up to 40% improvement in the effectiveness of the heat exchanger.

Keywords: Rapid cooling, Heat dissipation, linear flow, shell and tube heat exchanger, effectiveness.

Nome	Nomenclature				
Mfc	Mass flow rate of cold fluid [kg]	Mfh	Mass flow rate of hot fluid [kg]		
Tci	Cold fluid inlet temperature [°c]	Thi	Hot fluid inlet temperature [°c]		
Тсо	Cold fluid outlet temperature [°c]	Tho	Hot fluid outlet temperature [°c]		
Ср	Specific heat [kl/kg °c]	Di	Shell inner diameter [m]		
L	Total length of heat exchanger [m]	di	Tube inner diameter[m]		
do	Tube outer diameter [m]	f	Friction factor		
Nu	Nusset number	Re	Reynolds number		
ρ	Density [kg/m <sup>3</sup> ]	μ	Kinetic viscosity [kg/m sec]		
nf	Nanofluid	bf	Base fluid		
θm	Logarithmic mean temperature difference	Tm	Mean temperature [°c]		
Φ	Volume concentration	K	Thermal conductivity [W/m °c]		
Pr	Prantl number				

#### I. INTRODUCTION

Heat exchangers are the devices that transfer thermal energy or enthalpy from one medium to another, such that it solves the purpose of cooling. They are widely used now a days as condensers, radiators, Heat recovery units etc. there is no external heat and work transfer but happens only internally. The design of heat exchanger could be explained as, two fluids which transfer heat from one to another will be separated by a wall called heat transfer surface. Thus conduction and convection takes place. The fluids are at different temperatures, so temperature gradient occurs. Heat transfer surface is conducting surface and also there is a insulating material used to avoid external exchange of heat.

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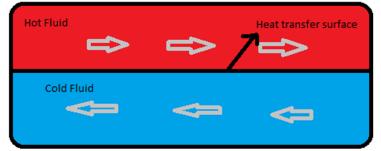


Fig 1. Process in heat exchangers

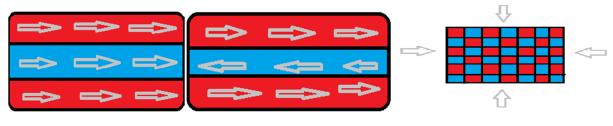
There are many types of heat exchangers generally used as per requirement. So we can classify them on basis of different parameters like flow direction, compactness of body, transfer type and construction. But we have a general classification of

Parallel flow Heat exchangers

Counter flow Heat exchangers

Cross flow Heat exchangers

Parallel flow heat exchangers are one in which two fluids flow in parallel to each other, In counter flow heat exchangers the fluids flow in opposite direction and in counter flow heat exchangers the flow occurs in combination of parallel and counter flow.



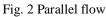


Fig. 3 Counter flow



In the present experiment we are using shell and tube heat exchanger for analysis. Because they are most versatile in process industries, in conventional and nuclear power stations, steam generators etc. These provide large ratio of surface area to volume. Shell & Tube heat exchangers are built of one shell and many number of tubes. And we make use of both parallel flow as well counter flow.

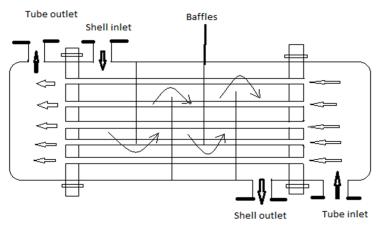


Fig. 5 Cross sectional view of shell and tube heat exchanger

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Basically heat exchangers use water, ethylene glycol etc. But here in this analysis we use nanofluids for better rate of heat transfer. Nanofluids are maiden used by Choi (1995)[1], then later many have experimented like L.B. Mapa, Sana Mazhar (2005)[2], "Heat transfer in mini heat exchanger using nanofluids", A.K. Singh (2008) [3], "Thermal conductivity of nanofluids", M.G. Mousa (2011)[6], "Effect of nanofluid concentration on the performance of circular heat pipe" and Xie et al (2011)[7], "Discussion on the thermal conductivity enhancement of nanofluids". All these experiments were conducted and tested with copper and aluminium nanoparticles. Usually any metal will have most enriched properties as a Power in Nano (10<sup>-9</sup>) size than in its solid form. So we are opting for nanoparticles of Magnesium oxide mixed with water thus making a nanofluid as a medium in heat exchanger. There are two basic steps for preparation of nanofluids. i.e., Single step method and Two step method. Single step method is one in which preparation and dispersion are done at a time. Two step method preparation and dispersion are done differently. Two step method is widely used because it can be used in mass production, so we are using two step method for our experiment. We here are purchasing Nano powders and dispersing them in water (basic fluid) using magnetic stirrer. This nanofluid is used in heat exchanger on cold water side.

#### II. EXPERIMENT

To make use of nanofluids we have to know the volume concentration we are using. So here we consider MgO Nanofluid with a volume concentration of 0.1% and 0.2%. For this we use the formula as follows

$$\Phi = \frac{\text{volume of nano particle}}{\text{volume of nano particle+volume of water}} * 100 \dots (1)$$

We take 30lit of water for experiment. So based on the above formulae we have calculated 108.36g & 207.48g respectively for 0.1% & 0.2% of MgO nanofluid to be dispersed.

We will first make the experiment only with de-ionized water for both parallel flow and counter flow. Then we will replace the water with the nanofluid once with 0.1% & then with 0.2 % MgO. For above three mediums we will do both counter flow and parallel flow. Thus we will note down the readings of each in every practical done. The arrangement of the experiment is shown in fig.6.



Fig. 6 Apparatus of Shell and Tube heat exchanger

In the above figure it is shown that the water tank is kept at some height to make gravitational flow constant. Then a water heater of capacity 1.5KW is used to heat the water. Then 4no's thermometers are used at each entrance and exit. These help in noting the readings at each section.

1	Material used in heat exchanger	Stainless Steel
2	Tube length	38.75cm
3	Number of tubes	9 no's
4	Tube inside diameter (di)	1.6cm
5	Tube outside diameter (do)	1.8cm
6	Shell inside diameter (Di)	10cm
7	Total heat exchanger length	60cm

Table 1: Specifications of heat exchanger

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Table 2	
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Properties of Fluids For Water			
Density,p	996.5Kg/m <sup>3</sup>		
Specific heat, Cp	4.181 KJ/Kg-K		
Thermal conductivity (K)	0.613W/mK		

Table 3

For Magnesium oxide (MgO):

Density,p	3580 Kg/m <sup>3</sup>
Specific heat, Cp	1030 KJ/Kg-K
Thermal conductivity (K)	60W/mK

#### III. CALCULATION STEPS FOR EXPERIMENTAL EVALUATION

For analysing the heat transfer coefficient we need to calculate some parameters. And also Reynolds number for knowing the flow charecteristics. So following are the Formulae to be calculated.

A. For Water Alone

At the mean temperature we have to take all the parameters like Density (p), Specific heat (Cp), Kinetic viscosity (µ), Prantl number (Pr) and thermal conductivity (K). With all these parameters we will find Reynolds number, nusset number, heat transfer coefficient, overall heat transfer.

$$\Theta m = LMTD = \frac{(Thi - Tci) - (Tho - Tco)}{ln \frac{(Thi - Tci)}{(Tho - Tco)}} \dots \dots \dots (2)$$

For parallel flow

$$\Theta m = LMTD = \frac{(Thi - Tco) - (Tho - Tci)}{ln \frac{(Thi - Tco)}{(Tho - Tci)}} \dots (3)$$

For counter flow

Nu = 
$$0.023^{*} \text{ Re}^{0.8_{*}} \text{ Pr}^{0.2}$$
 .....(5)

$$Nu = \frac{h*a}{k} \tag{6}$$

0.8

Effectiveness ( $\varepsilon$ ) =  $\frac{Ch (Thi-Tho)}{Cmin (Thi-Tci)}$ .....(9) Heat capacity (Q) =  $UA\theta m$  .....(10)

#### B. For Nanofluids

Density 
$$(\rho_{nf}) = (1-\Phi)\rho f + \Phi \rho p$$
 .....(11)

$$Cp = \frac{(1-\phi)*\rho f*Cpf + \phi*\rho p*Cpp}{\rho nf} \dots (12)$$

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#### IV. RESULTS AND DISCUSSIONS

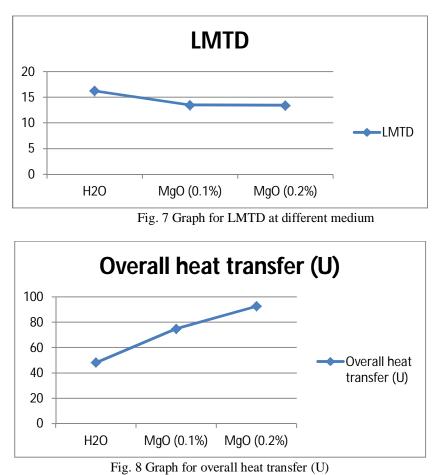
For making of nanofluids we need to use volume concentration formula (1), for finding the quantity of water taken. Since we have taken 30lits of water, we should calculate the amount of MgO nanoparticles used

Case (I): Concentration of 0.1%, we need to add 107.93gms of MgO Nanoparticles

Case (II) : Concentration of 0.2%, we need to add 216.09gms of MgO Nanoparticles

Now let us compare and conclude by using graphs. As we have used two flow rates of 1.2kg/min and 0.852kg/min we will have comparison for these two flow rates too

First of all we will look into the comparision of logarithmic mean temperature differences (LMTD), overall heat transfer (U) and effectiveness ( $\epsilon$ ) with each medium under the flow rate of 1.2kg/min and parallel flow at temperature of 50°C.



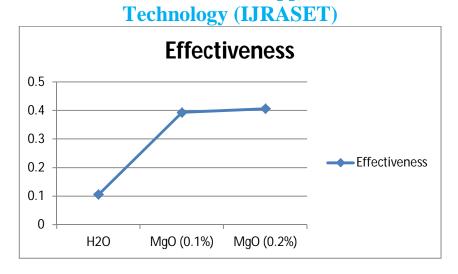


Fig. 9 Graph for Effectiveness

From the above graphs we can note that LMTD decreases when overall heat transfer and effectiveness increase drastically from water to MgO nanofluid.

Now we shall compare the same parameters for counter flow at 50°C and the flow rate of 1.2kg/min

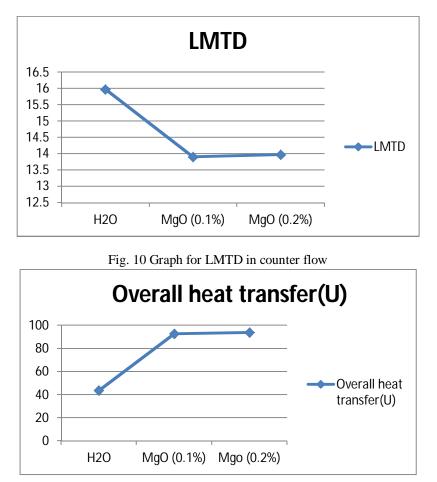


Fig.11 Graph for overall heat transfer (U)

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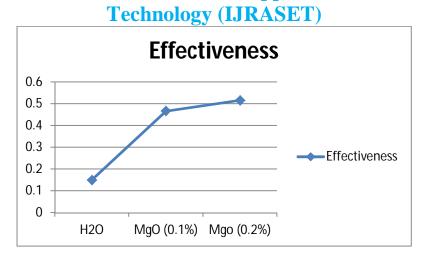


Fig.12 Graph for effectiveness in counter flow

Here also we see the same results of parallel flow.

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Then we shall change the flow rate now to 0.852kg/min, parallel flow at 50°C

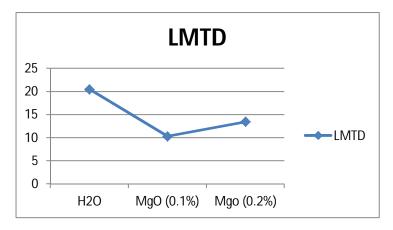


Fig. 13 Graph for LMTD at massflow rate of  $M_f B$ 

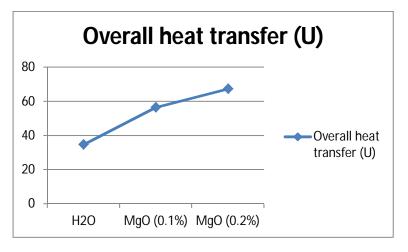
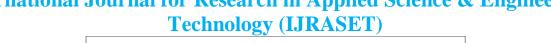


Fig. 14 Graph for U at mass flow rate  $M_f B$ 



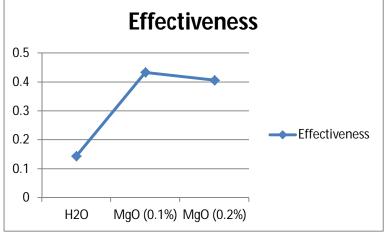


Fig.15 Graph for  $\epsilon$  at flow rate  $M_f B$ 

Next is counter flow in the changed flow rate at 50°C.

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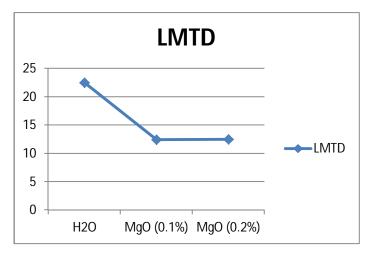


Fig.16 Graph of LMTD in counter flow

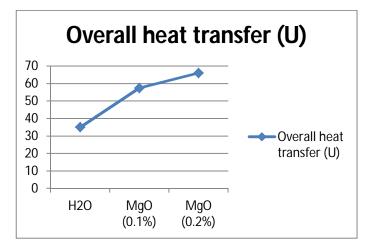


Fig. 17 Graph for U in counter flow

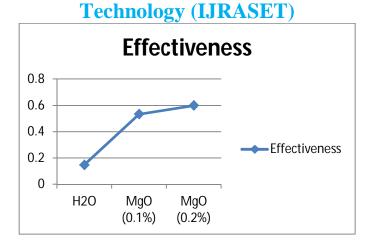


Fig.18 Graph for  $\varepsilon$  in counter flow

#### V. CONCLUSION

From the results obtained as per the experimental analysis, the overall heat transfer and effectiveness increases with decrease of LMTD. Highest effectiveness is obtained at the low flow rate and counter flow directions of fluid. Up to 40% of increment is seen in effectiveness of MgO nanofluid with 0.2% of MgO nanoparticle dispersion. Thus our assumption of increment of heat transfer rate has been proven.

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