



iJRASET

International Journal For Research in
Applied Science and Engineering Technology



INTERNATIONAL JOURNAL FOR RESEARCH

IN APPLIED SCIENCE & ENGINEERING TECHNOLOGY

Volume: 9 Issue: V Month of publication: May 2021

DOI: <https://doi.org/10.22214/ijraset.2021.34827>

www.ijraset.com

Call:  08813907089

E-mail ID: ijraset@gmail.com

Treatment of Alkaline Water using Reverse Osmosis System with the Help of Renewable Solar Energy

S. Dhinakaran¹, Bharathidasan²

¹PG Student, M. Tech Environmental Engineering, Prist University, Puducherry Campus, India.

²Head of the Department, Department of Civil Engineering, Prist University, Puducherry Campus, India.

Abstract: *The aim of this project is to develop a low-cost water purification system that provides clean drinking water. The project has a broad reach since drinking water pollution is a big global problem. This device would provide a solution to contaminated drinking water. Electric current is used to power the purifier during the water treatment process, reducing fluoride, chloride, and increasing the pH content of salty water, resulting in potable drinking water. To achieve this aim, a project management plan has been devised that divides the tasks amongst the community. The ultimate goal of designing this project is to provide clean drinking water at a low cost. The aim of this project is to develop and build a small-scale water purification system that needs little maintenance and is inexpensive. Initially, solar power would be used to power the purification system and to transition the system to renewable energy.*

Keywords: *Water Treatment, Reverse osmosis System, Water purification, Renewable energy.*

I. INTRODUCTION

In developing countries such as India, population growth combined with industrialization is putting a strain on water supply, causing the water table to shrink. Even in India's water-rich regions, the water table is receding. Water and fresh air are unavoidable life tools. Initially, solar power would be used to power the purification system and to transition the system to renewable energy. Daily monitoring of drinking water is the most important criterion for ensuring that everybody has access to safe drinking water. The testing results must be validated using standard values.

One of the access points that provides water quality reports and other insightful reports based on water collected at the locality level (lower than village level) in various parts of India. We have been regularly educating people in the villages on water testing methods using water testing kits for the past few years. To avoid problems and ensure proper operation, the water circulating in a boiler or cooling system must be continuously corrected. Water monitoring must be done on a regular basis to monitor impurities, treatment levels, degradation, and deposits. The aim of this project is to develop and build a small-scale water purification system that needs little maintenance and is inexpensive.

II. LITERATURE REVIEW

E.R. Cornelissen et al (2021) found that A minimal pre-treated reverse osmosis system using only screens on surface water did not result in stable reverse osmosis operation, however, the ultrafiltration pre-treated reverse osmosis system also failed in the longer run. Periodical air water cleaning in minimal pre-treated RO resulted in an approximately 4-fold reduction of the spacer clogging rate, while not affecting the membrane fouling rate.

Linyan Yang et al (2020) found that The Fe-based heterogeneous catalyst carried the higher potential to improve the biodegradability of ROC (i.e., 0.32 v. s. 0.27 for B/C, the ratio between BOD₅ and COD) although its direct COD removal efficiency was inferior to the homogeneous one (i.e., 49% v. s. 59% after 25 min' reaction). Jacob F. King et al (2020) found that Addition of 300 mg/L CaO at pH 11 achieved partial removal of the native nickel and copper by precipitation. Ozone pretreatment further enhanced precipitation of nickel, but not copper. Ozonation achieved 5-log inactivation of MS2 in all five concentrate samples at 1.18 mg O₃/mg DOC. Ozonation at 0.9 mg O₃/mg DOC formed 139–451 mg/L bromate. Pretreatment of RO concentrates with chlorine and ammonia reduced bromate formation by a maximum of 48% but increased total halogenated DBP concentrations from 20 µg/L to 36 µg/L. Regardless, neither bromate nor trihalomethane concentrations exceeded threshold concentrations of concern for discharge to marine waters.

III. WATER TESTING PROCESS

The goal of sampling is to obtain for analysis a portion of the main body of water that is truly representative. The most critical factors necessary to achieve this are:

- 1) Point(s) of sampling
- 2) Time of sampling
- 3) Frequency of sampling
- 4) Maintenance or integrity of the sample prior to analysis

A. Sampling

First of all, the testing team goes to the water body present in the vicinity to collect sample water to be tested. There could be various sources viz. Hand washing taps, Open Well, Tank or say underground water reservoir etc.



Figure 1: Sampling from hand wash Pipe



Figure 2: Sampling from tank



Figure 3: Sampling from Open well

IV. EXPERIMENTAL INVESTIGATION

A. Physical Test

The physical test involves in testing Taste (Using your tongue), Color (By looking at the color of water), Odour (By smelling the sample water), Transparency / Turbidity (By looking for the clarity of water) and Temperature (measured using thermometer)



Figure 4: Looking Colour of water

B. Chemical Test

1) pH

- The Ranges of PH Value for OPEN WELL WATER = 7.2
- The Ranges of PH Value for TANK WATER = 8.4
- The Ranges of PH Value for HAND WASHING PIPE WATER = 8.2

2) *Chlorides*: The Amount of chloride iron present in the whole of the given water sample = 0.118 gm/mg/ppm

3) Total hardness

- The Amount of total hardness of the given sample of water = 1800 ppm
- The Amount of permanent hardness of the given sample of water = 1750 ppm
- The Amount of temporary hardness of the given sample of water = 50 ppm

4) Fluoride

- The amount of Fluoride present in Open Well Water = 2.665 mg/liter
- The amount of Fluoride present in Tank Water = 3.743 mg/liter
- The amount of Fluoride present in Hand Washing Tap Water = 3.811 mg/liter

5) Total Dissolved Solids

- The TDS (Total Dissolved Solids) value for Open Well Water = 1450
- The TDS (Total Dissolved Solids) value for Tank Water = 1750
- The TDS (Total Dissolved Solids) value for Hand Washing Tap Water = 1980

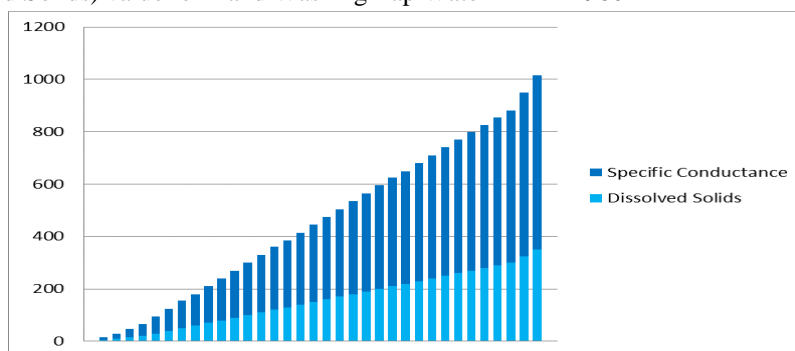


Figure 5: Dissolved Solids (ppm) Vs. Specific Conductance of Water

6) Alkalinity

- The Amount of HCO₃⁻-alkalinity present in Open Well Water = 675 ppm.
- The Amount of HCO₃⁻-alkalinity present in Hand Washing Tap Water = 850 ppm.
- The Amount of HCO₃⁻-alkalinity present in Tank Water = 725 ppm

V. WATER TREATMENT UNIT MODEL

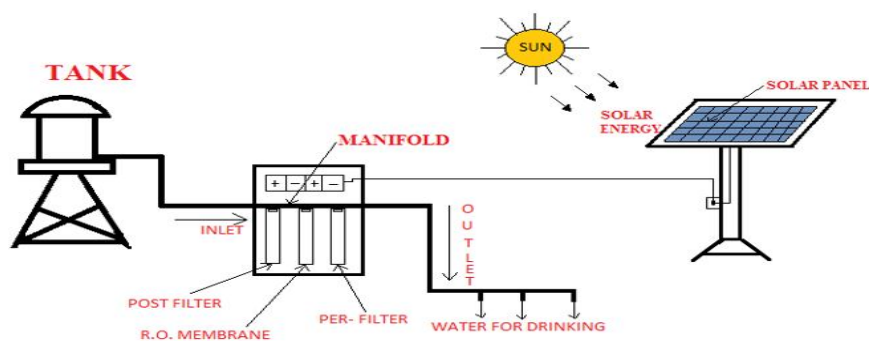


Figure 6: Treatment Unit with renewable energy

A. Solar Panel

The solar collector of the six units of 185watts was store in battery, the sunlight flow through the absorber. The absorber was a 2 mm temperature-resistant black porous felt mat where through the wire was stored as shown in figure 7.



Figure 6: Solar panel

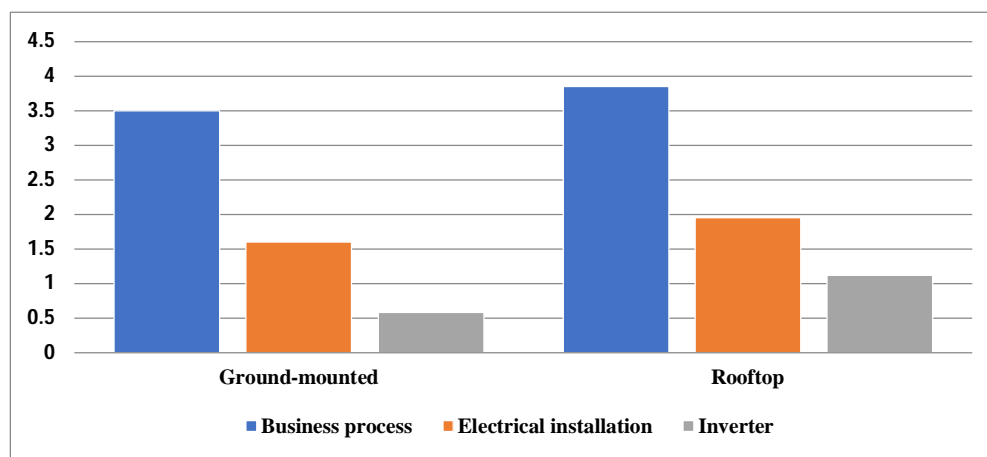


Figure 7: Solar Usage

B. Storage Tank

The storage tank capacity of more than 2500litres per day, were used to treated every day to stored.



Figure 8: Storage Tank

C. Reverse Osmosis (RO) Water Purification Method

Large-scale reverse osmosis water purification filters and facilities play an important role in turning salt water to fresh drinking water in many countries. The procedure is extremely simple. Bore water reaches high-pressure pumps before passing through a series of membranes that filter out salt water and other unwanted particles.



Figure 9: Water Purifier

VI. CONCLUSIONS

The following conclusions were drawn at the end of this study

- A. The project was a success, and it was able to refine the design of a cost-effective, but reliable, easy-to-manufacture, and simple-to-use water purification filter for providing a balance between the elements of people, prosperity, and the earth.
- B. The most common bacterial pathogens found in water around the world. The current configuration filters approximately 0.4-1 L/hr and uses approximately 5mL of 2000-ppm colloidal silver per filtration device. Because of its enclosed nature, the new device poses no risk of leakage or contamination of filtered water.
- C. In comparison to other methods of water treatment, our project is to treated groundwater recharge is ecologically safe and less expensive than chemically induced coagulation, relying solely on reverse osmosis and solar energy to power the purifier.

REFERENCES

- [1] E.R. Cornelissen, D.J.H. Harmsen, B. Blankert, L.P. Wessels, W.G.J. van der Meer, Effect of minimal pre-treatment on reverse osmosis using surface water as a source, *Desalination*, Volume 509, 2021, 115056, ISSN 0011-9164, <https://doi.org/10.1016/j.desal.2021.115056>.
- [2] Linyan Yang, Mei Sheng, Yejin Li, Weibo Xue, Kai Li, Guomin Cao, A hybrid process of Fe-based catalytic ozonation and biodegradation for the treatment of industrial wastewater reverse osmosis concentrate, *Chemosphere*, Volume 238, 2020, 124639, ISSN 0045-6535, <https://doi.org/10.1016/j.chemosphere.2019.124639>.
- [3] Jacob F. King, Aleksandra Szczuka, Zhong Zhang, William A. Mitch, Efficacy of ozone for removal of pesticides, metals and indicator virus from reverse osmosis concentrates generated during potable reuse of municipal wastewaters, *Water Research*, Volume 176, 2020, 115744, ISSN 0043-1354, <https://doi.org/10.1016/j.watres.2020.115744>.
- [4] A. Mavhungu, V. Masindi, S. Foteinis, R. Mbaya, M. Tekere, I. Kortidis, E. Chatzisyneon, Advocating circular economy in wastewater treatment: Struvite formation and drinking water reclamation from real municipal effluents, *Journal of Environmental Chemical Engineering*, Volume 8, Issue 4, 2020, 103957, ISSN 2213-3437, <https://doi.org/10.1016/j.jece.2020.103957>.
- [5] Meihong Liu, Qingyuan He, Zhongwei Guo, Kaifei Zhang, Sanchuan Yu, Congjie Gao, Composite reverse osmosis membrane with a selective separation layer of double-layer structure for enhanced desalination, anti-fouling and durability properties, *Desalination*, Volume 499, 2021, 114838, ISSN 0011-9164, <https://doi.org/10.1016/j.desal.2020.114838>.
- [6] Jennifer Hooper, Denise Funk, Kati Bell, Morayo Noibi, Kyle Vickstrom, Chris Schulz, Eddie Machek, Ching-Hua Huang, Pilot testing of direct and indirect potable water reuse using multi-stage ozone-biofiltration without reverse osmosis, *Water Research*, Volume 169, 2020, 115178, ISSN 0043-1354, <https://doi.org/10.1016/j.watres.2019.115178>.
- [7] Aditya Sharma, Bhuvneshwar Agnihotri, Sankalp Vemavarapu, Akhilendra B. Gupta, Chemistry of inorganic scaling in full-scale reverse osmosis plants treating brackish groundwater, *Journal of Environmental Chemical Engineering*, Volume 8, Issue 5, 2020, 104108, ISSN 2213-3437, <https://doi.org/10.1016/j.jece.2020.104108>.
- [8] Bongyeon Jung, Caroline Y. Kim, Shiyuan Jiao, Unnati Rao, Alexander V. Dudchenko, Jefferson Tester, David Jassby, Enhancing boron rejection on electrically conducting reverse osmosis membranes through local electrochemical pH modification, *Desalination*, Volume 476, 2020, 114212, ISSN 0011-9164, <https://doi.org/10.1016/j.desal.2019.114212>.

- [9] Caleb Hirsimaki, John G. Outram, Graeme J. Millar, Ali Altaee, Process simulation of high pH reverse osmosis systems to facilitate reuse of coal seam gas associated water, *Journal of Environmental Chemical Engineering*, Volume 8, Issue 5, 2020, 104122, ISSN 2213-3437, <https://doi.org/10.1016/j.jece.2020.104122>.
- [10] Li Gao, Gang Liu, Arash Zamyadi, Qilin Wang, Ming Li, Life-cycle cost analysis of a hybrid algae-based biological desalination – low pressure reverse osmosis system, *Water Research*, Volume 195, 2021, 116957, ISSN 0043-1354, <https://doi.org/10.1016/j.watres.2021.116957>.
- [11] Yufang Li, Mengchen Li, Kang Xiao, Xia Huang, Reverse osmosis membrane autopsy in coal chemical wastewater treatment: Evidences of spatially heterogeneous fouling and organic-inorganic synergistic effect, *Journal of Cleaner Production*, Volume 246, 2020, 118964, ISSN 0959-6526, <https://doi.org/10.1016/j.jclepro.2019>.
- [12] Thiago Santos de Almeida Lopes, Rainer Heßler, Christoph Bohner, Gilson Barbosa Athayde Junior, Rennio Felix de Sena, Pesticides removal from industrial wastewater by a membrane bioreactor and post-treatment with either activated carbon, reverse osmosis or ozonation, *Journal of Environmental Chemical Engineering*, Volume 8, Issue 6, 2020, 104538, ISSN 2213-3437, <https://doi.org/10.1016/j.jece.2020.104538>.
- [13] I. Fuoco, A. Figoli, A. Criscuoli, G. Brozzo, R. De Rosa, B. Gabriele, C. Apollaro, Geochemical modeling of chromium release in natural waters and treatment by RO/NF membrane processes, *Chemosphere*, Volume 254, 2020, 126696, ISSN 0045-6535, <https://doi.org/10.1016/j.chemosphere.2020.126696>.
- [14] Amaia Lejarazu-Larrañaga, Serena Molina, Juan Manuel Ortiz, Gerardo Riccardelli, Eloy García-Calvo, Influence of acid/base activation treatment in the performance of recycled electromembrane for fresh water production by electrodialysis, *Chemosphere*, Volume 248, 2020, 126027, ISSN 0045-6535, <https://doi.org/10.1016/j.chemosphere.2020.126027>.
- [15] Ruth Habte Hailemariam, Yun Chul Woo, Mekdimu Mezemir Damtie, Bong Chul Kim, Kwang-Duck Park, June-Seok Choi, Reverse osmosis membrane fabrication and modification technologies and future trends: A review, *Advances in Colloid and Interface Science*, Volume 276, 2020, 102100, ISSN 0001-8686, <https://doi.org/10.1016/j.cis.2019.102100>.
- [16] Zhewei Zhang, Omkar R. Lokare, Alen V. Gusa, Radisav D. Vidic, Pretreatment of brackish water reverse osmosis (BWRO) concentrate to enhance water recovery in inland desalination plants by direct contact membrane distillation (DCMD), *Desalination*, Volume 508, 2021, 115050, ISSN 0011-9164, <https://doi.org/10.1016/j.desal.2021.115050>.
- [17] Xu Ren, Kai Song, Yu Xiao, Shaoyan Zong, Dan Liu, Effective treatment of spacer tube reverse osmosis membrane concentrated leachate from an incineration power plant using coagulation coupled with electrochemical treatment processes, *Chemosphere*, Volume 244, 2020, 125479, ISSN 0045-6535, <https://doi.org/10.1016/j.chemosphere.2019.125479>.
- [18] Karin Kiefer, Tobias Bader, Nora Minas, Elisabeth Salhi, Elisabeth M.-L. Janssen, Urs von Gunten, Juliane Hollender, Chlorothalonil transformation products in drinking water resources: Widespread and challenging to abate, *Water Research*, Volume 183, 2020, 116066, ISSN 0043-1354, <https://doi.org/10.1016/j.watres.2020.116066>.
- [19] Xin Tong, Yong Cui, Yun-Hong Wang, Yuan Bai, Tong Yu, Xue-Hao Zhao, Nozomu Ikuno, Hui-jia Luo, Hong-Ying Hu, Yin-Hu Wu, Fouling properties of reverse osmosis membranes along the feed channel in an industrial-scale system for wastewater reclamation, *Science of The Total Environment*, Volume 713, 2020, 136673, ISSN 0048-9697, <https://doi.org/10.1016/j.scitotenv.2020.136673>.
- [20] Sebastian Franzsen, Craig Sheridan, Geoffrey S. Simate, Process for high recovery treatment of brackish water reverse osmosis concentrate, *Desalination*, Volume 498, 2021, 114792, ISSN 0011-9164, <https://doi.org/10.1016/j.desal.2020.114792>.
- [21] Vimeipha Vilayphone, John G. Outram, Fiona Collins, Graeme J. Millar, Ali Altaee, Process design of coal seam gas associated water treatment plants to facilitate beneficial reuse, *Journal of Environmental Chemical Engineering*, Volume 8, Issue 5, 2020, 104255, ISSN 2213-3437, <https://doi.org/10.1016/j.jece.2020.104255>.
- [22] C. Cox and J. Graham, "Steps towards automatic clarification control," in *IEEE Colloquium on Advances in Control in the Process Industries: An Exercise in Technology Transfer*, pp. 6/1–6/4, March 1994.



10.22214/IJRASET



45.98



IMPACT FACTOR:
7.129



IMPACT FACTOR:
7.429



INTERNATIONAL JOURNAL FOR RESEARCH

IN APPLIED SCIENCE & ENGINEERING TECHNOLOGY

Call : 08813907089  (24*7 Support on Whatsapp)