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Experimental and Analysis of Aluminum –CU Composite with Different Composition

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Abstract-Copper-aluminum alloy or brass is known for their corrosion resistance. Brasses are stronger and more ductile than red and semi red brasses. They have high wear resistance and low friction coefficient against steel. The room temperature phase transformations are slow and usually do not occur; therefore these alloys are single phase alloys. The bronzes are used in bearings, gears, piston rings, valves and fittings. Aluminum is added to copper in order to improve machinability and toughness. Aluminum increases the tensile strength and ductility of the copper, but the composition can be adjusted to balance machinability and strength requirements. These alloys have a slow fail mechanism that temporarily prevents galling and seizing. In this project we have been added aluminum 10%, 20%, and 30% and which one is the best mechanical properties for automobile, aircraft component. In order that some experimental investigation conducting and finally concluded the best mechanical property in this composition.

Keywords-*Al-Cu composites, Hardness test, Impact test, Microstructure test, Electrical Resistivity test.*

I. INTRODUCTION

A. Introduction

Composite materials are materials made from two or more constituent materials with significantly different physical or chemical properties which remain separate and distinct on a macroscopic level within the finished structure. The greatest advantage of composite materials is strength and stiffness combined with lightness. By choosing an appropriate combination of reinforcement and matrix material, manufacturers can produce properties that exactly fit the requirements for a particular structure for a particular purpose.

II. METAL MATRIX COMPOSITE

A. Introduction

Metal matrix composite (MMC) is composite material with at least two constituent parts, one being a metal. The other material may be a different metal or another material, such as a ceramic or organic compound. MMCs are nearly always more expensive than the more conventional materials they are replacing. As a result, they are found where improved properties and performance can justify the added cost. Today these applications are found most often in aircraft components, space systems and high-end or —boutique sports equipment. The scope of applications will certainly increase as manufacturing costs are reduced.

B. Reinforcement

The reinforcement material is embedded into the matrix. The reinforcement does not always serve a purely structural task (reinforcing the compound), but is also used to change physical properties such as wear resistance, friction coefficient, or thermal conductivity. The reinforcement can be either continuous, or discontinuous. Discontinuous MMCs can be isotropic, and can be worked with standard metalworking techniques, such as extrusion, forging or rolling. In addition, they may be machined using conventional techniques, but commonly would need the use of polycrystalline diamond tooling (PCD).

III. PROPERTIES OF MATRIX

A. General Properties Of Copper

1) General Properties

Name, symbol, number-copper, Cu, 29

Standard atomic weight-63.546g.mol⁻¹

Electron configuration-[Ar] 3d¹⁰ 4s¹

Electrons per shell 2, 8, 18, 1

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2) Physical Properties

Phase	solid
Density (near r.t.)	8.94 g·cm ⁻³
Liquid density atm.p.	8.02 g·cm ⁻³
Melting point	1357.77 K1084.62 °, C1984.32 °, F
Boiling point	2835 K2562 °, C4643 °, F
Heat of fusion	13.26 kJ·mol ⁻¹
Heat of vaporization	300.4 kJ·mol ⁻¹
Specific heat capacity	(25 °C) 24.440 J·mol ⁻¹ ·K ⁻¹

3) Atomic Properties

Oxidation states (Mildly basic oxide)	+1, +2, +3, +4
Electro negativity	1.90 (Pauling scale)
Ionization energies (More)	1st: 745.5 kJ·mol ⁻¹ 2nd: 1957.9 kJ·mol ⁻¹ 3rd: 3555 kJ·mol ⁻¹
Atomic radius	128 pm
Covalent radius	132±4 pm
Van der Waals radius	140 pm

4) Miscellanea

Crystal structure	face-centered cubic
Magnetic ordering	diamagnetic
Electrical Resistivity	(20 °C) 16.78 nΩ·m
Thermal conductivity	(300 K) 401 W·m ⁻¹ ·K ⁻¹
Thermal expansion	(25 °C) 16.5 μm·m ⁻¹ ·K ⁻¹
Speed of sound(thin rod) (r.t.) (annealed)	3810 m·s ⁻¹
Young's modulus	110–128 GPa
Shear modulus	48 GPa
Bulk modulus	140 GPa
Poisson ratio	0.34
Mohs hardness	3.0
Vickers hardness	369 MPa
Brinell hardness	874 MPa
CAS registry Number	7440-50-8

5) Most Stable Isotopes

Main article: Isotopes of copper

iso	NA	half-life	DM	DE (MeV)	DP
63	Cu	69.15%		⁶³ Cu is stable with	34 neutrons

B. General Properties of Aluminium

1) General Properties

Name, symbol, number- aluminium, Al, 13
 Standard atomic weight-26.9815386g·mol⁻¹

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Electron configuration-[Ne] 3s² 3p¹
 Electrons per shell 2, 8, 3

2) Physical Properties

Phase	solid
Density (near r.t.)	2.70 g·cm ⁻³
Liquid density atm.p.	2.375 g·cm ⁻³
Melting point	933.47 K 660.32 °, C 1220.58 °, F
Boiling point	2792 K 2519 °, C 4566 °, F
Heat of fusion	10.71 kJ·mol ⁻¹
Heat of vaporization	294.0 kJ·mol ⁻¹
Specific heat capacity	(25 °C) 24.200 J·mol ⁻¹ ·K ⁻¹

3) Atomic Properties

Oxidation states	3,2,1
(Mildly basic oxide)	
Electro negativity	1.61 (Pauling scale)
Ionization energies	
(More)	1st: 577.5 kJ·mol ⁻¹ 2nd: 1816.7 kJ·mol ⁻¹ 3rd: 2744.8 kJ·mol ⁻¹
Atomic radius	143 pm
Covalent radius	121±4 pm
Van der Waals radius	184 pm

4) Miscellanea

Crystal structure	face-centered cubic
Magnetic ordering	paramagnetic
Electrical Resistivity	(20 °C) 28.2 nΩ·m
Thermal conductivity	(300 K) 237 W·m ⁻¹ ·K ⁻¹
Thermal expansion	(25 °C) 23.1 μm·m ⁻¹ ·K ⁻¹
Speed of sound(thin rod)	
(r.t.) (annealed)	5000 m·s ⁻¹
Young's modulus	70 GPa
Shear modulus	26 GPa
Bulk modulus	76 GPa
Poisson ratio	0.35
Mohs hardness	2.75
Vickers hardness	167 MPa
Brinell hardness	245 MPa
CAS registry	
Number	7429-90-5

5) Most Stable Isotopes

Main article: Isotopes of copper

iso	NA	half-life	DM	DE (MeV)	DP
²⁷ Al	100	²⁷ Al is stable with 14 neutrons			

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IV. MATERIAL REQUIREMENT FOR VARIOUS RATIOS

Requirement size -100 x 100 x18
 Volume=10x10x1.8=180cm³

A. Mixing ratio-90%&10% (Copper and Aluminium)

Volume of copper=180*0.90=162 gm
 Volume of Aluminum=180*.10=18 gm
 Density of copper=8.9gm/cc
 Weight of copper 162*8.9=1441.8 gm =1.441kg
 Density of aluminum=2.7gm/cc
 Weight of Aluminum =18x2.7= 48.6 gm 30 % for excess (Runner&riser,slag)
 Total weight of mixture
 Copper=1441.8 gm +410 gm= 1851.8 gm
 Aluminum=48.6+15=63.6 gm

B. Mixing ratio-80%&20%(Copper and Aluminium)

Volume of copper=180*0.80=144 gm
 Volume of Aluminum=180*.20=36 gm
 Density of copper=8.9 gm/cc
 Weight of copper 144*8.9=1281.6 gm/cc=1.281kg
 Density of aluminum=2.7gm/cc
 Weight of Aluminum =36x2.7=97.2 gm
30 % for excess (Runner&riser, slag) Total weight of mixture
 Copper=1281.6Kg+360 g= 1641.6gm Aluminum=97.2+30=127.2 gm

C. Mixing ratio-70%&30%(Copper and Aluminium)

Volume of copper=180*0.70=126 gm Volume of Aluminum=180*.30=54 gm Density of copper=8.9gm/cc
 Weight of copper 126*8.9=1121.4 gm =1.121kg Density of aluminum=2.7gm/cc
 Weight of Aluminum =54x2.7=145.8 gm 30 % for excess
 (runner &riser, slag) Total weight of mixture
 Copper=1121.4g+300 g= 1421.4gm=1.421kg
 Aluminum=145.8+45=190.8gm

V. DESTRUCTIVE TEST

A. Rockwell Hardness Test

1) Hardness Details

TYPES: Rockwell Hardness
 Major Load Applied: 100Kgf
 Types of Indenter used: 1/16 —

2) Verification of Raw Materials

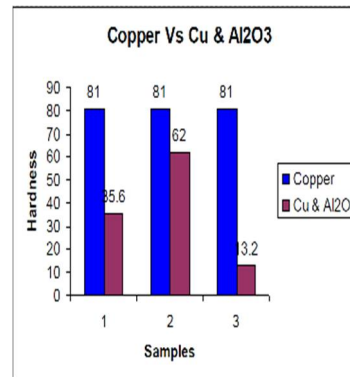
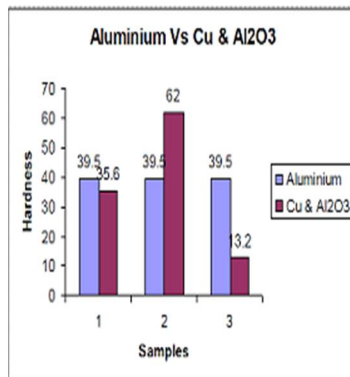
S.No	Material	Trial 1	Trial 2	Trial 3	Trial 4	Trial 5	Mean
1	Pure Aluminum(LM4)	39	41	48	40	40	39.5
2	Pure Copper	81	79	78	83	83	81

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3) Verification of Composite Materials

S.No	Material	Trial 1	Trial 2	Trial 3	Trial 4	Trial 5	Mean
1	Cu – 90% - Al ₂ O ₃ – 10%	35	39	33	32	39	35.6
2	Cu – 80% - Al ₂ O ₃ – 20%	54	58	60	69	68	62
3	Cu – 70% - Al ₂ O ₃ – 30%	12	16	14	13	11	13.2

4) Comparison Graph



Sample 1 : 90% Cu 10% Al₂O₃

Sample 2 : 80% Cu 20% Al₂O₃

Sample 3 : 70% Cu 30% Al₂O₃

B. Impact Test

a) Impact Strength

Izod method.

Specification of the Machine :

Energy Range = 0 – 168 J

Least Count (1 Division) = 2J

SPECIMEN LENGTH-75 mm

SIZE-10Sqmm

NOTCH- V NOTCH (45° Included Angle)

FALL ANGLE-90°

SPECIMEN SUPPORTING- Cantilever Beam setup

NOTCH DEPTH-2mm

AREA- a²

$$= (10 - 2)^2$$

$$= 64\text{mm}^2$$

COMPOSITION I:

90%-10% Cu and Al₂O₃

I = K/A J/m²

I = Impact Strength

K = Energy Observed

A = Area

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Energy Observed = 62J/mm²

= 62/64 = 0.9687 J/mm²

COMPOSITION II :

80%-20% Cu and Al₂O₃

I = K/A J/m²

Energy Observed = 74J/mm²

= 74/64 = 1.15 J/mm²

COMPOSITION III:

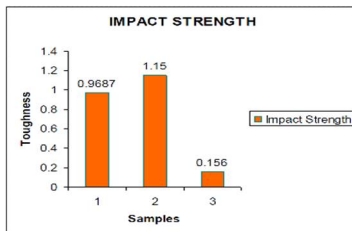
70%-30% Cu and Al₂O₃

I = K/A J/m²

Energy Observed = 10J/mm²

= 10/64 = 0.156 J/mm²

COMPARISON GRAPH



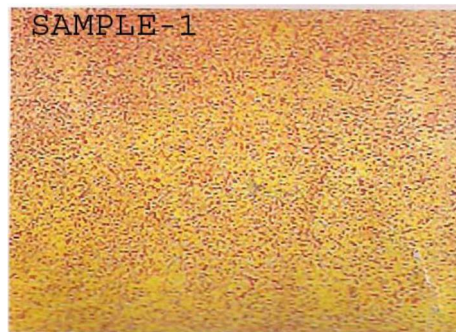
Sample 1 : 90% Cu 10% Al₂O₃

Sample 2 : 80% Cu 20% Al₂O₃

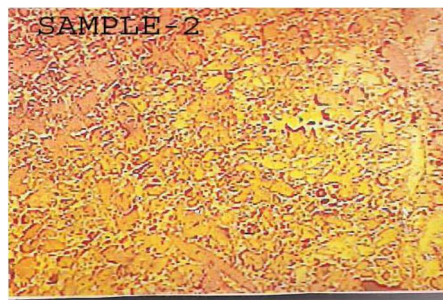
Sample 3 : 70% Cu 30% Al₂O₃

C. Microstructure

1) *Microstructure Results:* Microstructure of Cu-Al₂O₃ at 90%-10%



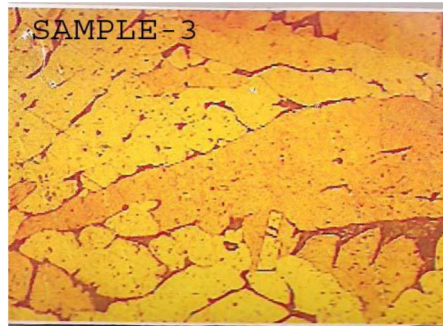
The matrix shows completely fine transformed beta as the matrix. Some equi-axed alpha also present in the matrix of beta.
Microstructure of Cu-Al₂O₃ at 80%-20%



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The matrix large grains of alpha in a matrix of beta solid solution. The matrix also shows the intergranular voids. This may also due to shrinkage defect taken place during casting.

Microstructure of Cu-Al₂O₃ at 70%-30%



The matrix shows cast fine inter-dendritic grains of alpha and beta. The matrix is beta and the presence of alpha solid solution is about 20% in a matrix of beta solid solution. Void shows the material is cast and due to shrinkage defect

D. Electrical Resistivity

Electrical Resistivity

S.NO	sample	size	Resistance
1	Cu-Al ₂ O ₃ at 90%-10%	90x90	0.3
2	Cu-Al ₂ O ₃ at 80%-20%	90x90	0.4
3	Cu-Al ₂ O ₃ at 70%-30%	90x90	0.2

Resistivity = $\frac{\text{Resistance} \times \text{Area}}{\text{Length}}$

Length

COMPOSITION I :

90%-10% Cu and Al₂O₃

$$= \frac{0.3 \times 1620}{90} = 5.4 \text{ Ohms}$$

COMPOSITION II :

90%-10% Cu and Al₂O₃

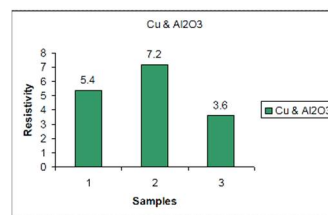
$$= \frac{0.4 \times 1620}{90} = 7.2 \text{ Ohms}$$

COMPOSITION III:

90%-10% Cu and Al₂O₃

$$= \frac{0.2 \times 1620}{90} = 3.6 \text{ Ohms}$$

COMPARISON GRAPH



VI. CONCLUSION

Composite materials especially Cu-Al₂O₃ composites having good mechanical properties compared with the conventional materials. It is used in various industrial application these materials having light weight along with high hardness .It with stand high load

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compare with the existing materials are most applicable in the engineering products instead of existing materials.

Finally I conclude that the percentage of Aluminum increases automatically the hardness and all the parameters are increasing

REFERENCES

- [1] H.E. Boyer, T.M. Gall, Metals Handbook, Desk Ed., American Society for Metals, Metals Park, OH, 1991, pp. 20.16–20.21.
- [2] G. Gusmano, A. Bianco, R. Polini, J. Mater. Sci. 36 (2001) 901–907.
- [3] J.L. Johnson, R.M. German, Int. J. Powder Metall. 30 (1) (1994) 91–100
- [4] W.F. Wang, Powder Metall. 40 (4) (1997) 295–300.
- [5] R. Jedamzik, A. Neubrand, J. Rodel, J. Mater. Sci. 35 (2000) 477– 486
- [6] T.H. Ihn, S.W. Lee, S.K. Joo, Powder Metallurgy, vol. 37, No. 4, 1994, pp. 283–288.
- [7] V.N. Eremenko, R.V. Minakova, M.M. Churakov, Sov. Powder Metall. Met. Ceram. 15
- [8] K. Byoong, Mechano-chemical process for production of high density and ultrafine W/Cu composite material, US Patent no. 5842108 (1998).
- [9] M.K. Yoo, Tungsten skeleton structure fabrication method employed in application of copper infiltration and W–Cu composite material fabrication method thereof, US Patent no. 5963773 (1999).
- [10] S.E. Allen, Proceedings of the 44th IEEE Conference on Electrical Contacts, 1998, 276–2 85



10.22214/IJRASET



45.98



IMPACT FACTOR:
7.129



IMPACT FACTOR:
7.429



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