

Synthesis of Nanoparticles Used For Solar Heating Purpose by Applying On Window

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Abstract: *The applications of passive heating techniques are increasing now these days. Further, the application of glass based windows and walls in the multistory buildings and complexes are also a new trend adopted by the building planners. For increasing the heat gain through glass windows/roofs different coatings have been analyzed in the present work. These coatings have been prepared by using different nanoparticles viz. magnesium oxide nanoparticle, copper oxide nanoparticle and aluminium oxide nanoparticle. It has been found that copper oxide based coating is best for increasing the heat gain.*

Keyword: *Passive cooling technique, Magnesium oxide nanoparticle, Copper oxide nanoparticle, Aluminium oxide nanoparticle, Glass and Thermocol.*

I. INTRODUCTION

A. Glass

Glass is a non-crystalline amorphous solid that is frequently transparent and has extensive practical and technological usage. The most well know and oldest types of glass are "silicate glasses" based on the chemical compound silica. Glass can transmit reflect and refract light, this type of properties of glass can be increase by cutting and polishing of glass. Engineering Properties of Glass are transparency, strength, workability, transmittance, U value and recycle property[1].

B. Solar radiation

Solar radiation is also known as short-wave radiation. Solar radiation comes in various forms such as visible light, radio waves, heat (infrared), x-rays and ultraviolet rays.

Availability of solar radiation on earth surface is higher on sunny day and usually low on cloudy days. The sun is a source of heat radiations and it emits radiations in all the directions. The atmosphere absorbs heat radiations and air, clouds, dust particles etc. present in the atmosphere spread out the heat and light radiations. Obviously, the earth receives only a small part of the energy emitted by the sun.

Much of the solar radiation is transmitted through the glass covering and absorbed by the objects within the enclosure. Most of the thermal radiation emitted at low temperature is reflected back and remains inside. Because of this one way action of heat exchange of the glass the temperature within the enclosure becomes considerably higher than the ambient temperature outside. The phenomenon is commonly referred to as the greenhouse effect. The sky creates a partial greenhouse effect if it is heavily loaded with CO₂, H₂O and to a lesser extent ozone[2].

C. Nanoparticles

Nanoparticles are particles of size between 1 and 100 nanometers. In nanotechnology, a particle is defined as a small object that conducts as a whole unit with respect to its transport and properties. Scientific research on nanoparticles is acute as they have many potential applications in medicine, physics, optics, and electronics. The term "nanoparticle" is not usually applied to individual molecules; it usually refers to inorganic materials. Nanoparticles can demonstrate size-related properties importantly different from those of fine particles. There are several synthesis methods for creating nanoparticles like physical vapor deposition, chemical vapor deposition, sol-gel method, RF plasma method, pulsed laser method, thermolysis and combustion method. Combustion synthesis is an effective ecofriendly and low-cost method for preparation of nanoparticles[3][4][5].

II. EXPERIMENTAL SET UP AND METHODOLOGY

A. Preparation of nanoparticles

There are various methods available for synthesis of nanoparticles such as physical vapor deposition (PVD), chemical vapor deposition (CVD), sol-gel method, RF plasma method, pulsed laser method, thermolysis and combustion method. In the present work nanoparticles are synthesis by using combustion method. The combustion method is an easier method to make oxides of any element by using its nitrate. Equipments used for this process are muffle furnace, mortar and pestle, nitrates (copper nitrate, aluminium nitrate and magnesium nitrate), urea, crucible, pyrometer and weigh machine.

Table 1 Properties of chemical which used

Property	Magnesium nitrate	Aluminium nitrate	Copper nitrate	Urea
Formula	$Mg(NO_3)_2$	$Al(NO_3)_3$	$Cu(NO_3)_2$	CH_4N_2O
Molar mass	148.3 g/mol	212.996g/mol	187.56 g/mol	60.06g/mol
Melting point	89 ⁰ C	72.8 ⁰ C	114.5 ⁰ C	133 ⁰ C
Density	2.3g/cm ³	1.72g/cm ³	3.05g/cm ³	1.32g/cm ³

For the synthesis of nanoparticles by using combustion method initially the ratio between nitrates and urea has been calculated subsequently urea and nitrate are mixed in mortar to form a gel. The process of crushing and mixing of urea and nitrates took 5 to 6 minutes to form gel like solution. About 3 to 4 ml of distilled water is also added into the mixture of urea and nitrate during the formation of gel. The process of mixing results an exothermic reaction and small amount of heat is liberated when urea is added within it. After formation of gel like solution it has been placed into crucible and then crucible is placed into muffle furnace where temperature has been maintained around 550⁰C.



Figure 1 Muffle furnace at 550⁰C

Hot fumes with reddish flame come out of cup for 10-12 seconds and then the gel is converted into the oxides. With the help of tongue cup is taken out form the furnace and keep for 20 to 30 minute to cool.

B. Testing zone

The testing setup consists of ten cubical shape structure made up of glass and polystyrene (Thermocol). Equipments used for this process are glass, polystyrene (thermocool), thermocouple, water bowl, adhesive (gum), Nanoparticles (magnesium oxide, aluminium oxide and copper oxide) and weigh machine.



Figure 2 Experimental Setup

In the above figure glasses are coated with the mixture/paste of copper oxide nanoparticle and adhesive

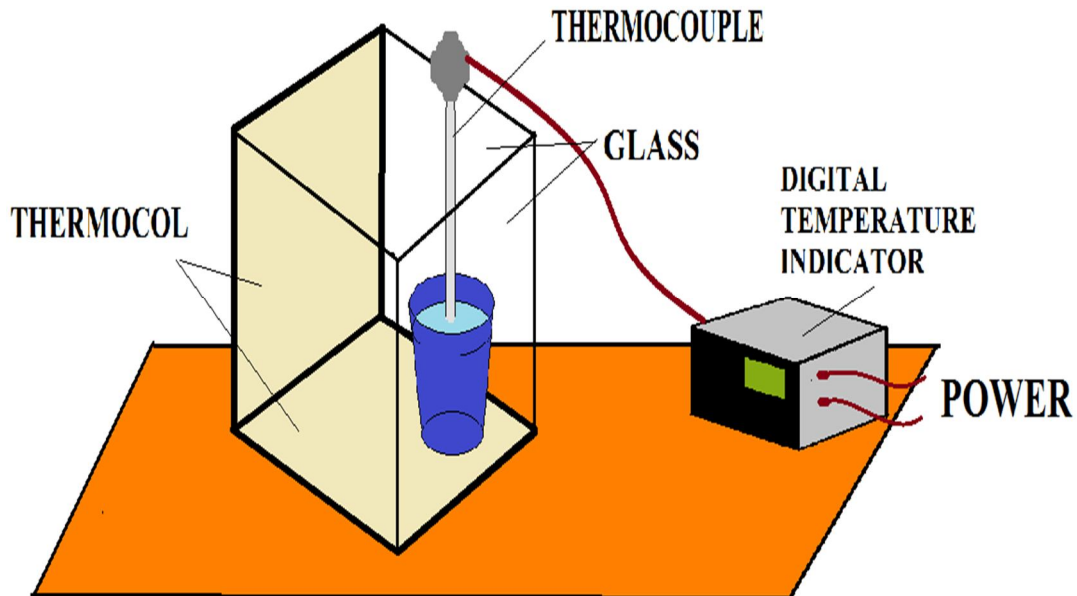


Figure 3 Schematic diagram

Ten different cubical structures of size $0.2 \times 0.2 \times 0.2 \text{ m}^3$ have been made for performing the experiments. These cubes are having four faces of glass and other two faces are of polystyrene (thermocol). All the cubical structures are made in similar manner as described above. Three cubical structures are coated with aluminium oxide nanoparticles using 5gm, 10gm and 15gm with the help of adhesive on the glass faces. Three cubical structures are coated with copper oxide nanoparticles using 5gm, 10gm and 15gm with the help of adhesive on the glass faces. Three cubical structures are coated with magnesium oxide nanoparticles using 5gm, 10gm and 15gm with the help of adhesive on the glass faces and one cubical structure is coated with only adhesive i.e. without nanoparticles on the glass faces. A bowl of water is kept inside the cubical structure setup and thermocouple is placed to measure temperature of water. All the setup is kept in sunlight for 5-6 hrs.

III. RESULTS AND DISCUSSIONS

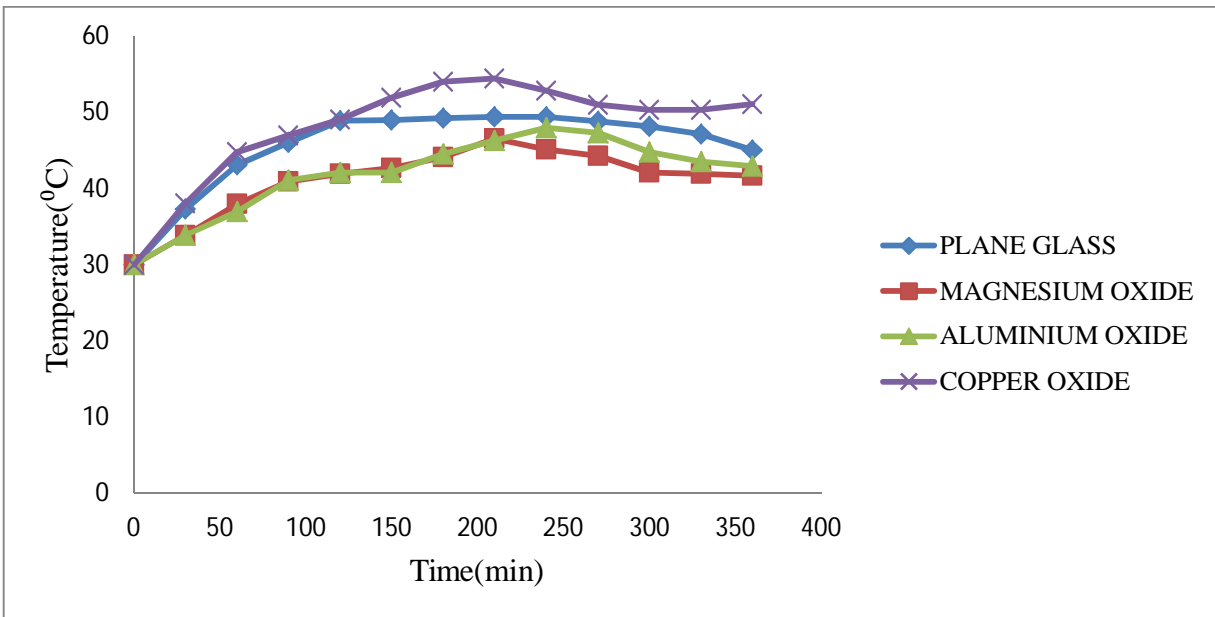


Figure 4 Comparison between time and temperature for 5gm nanoparticles

In the present work different types of nanoparticles have been prepared and applied over the test zone. The comparisons of solar heat gain in different conditions have been studied.

When 5gm nanoparticles are mixed with adhesive and coated on the glass faces-

- a) Plane glass: Maximum temperature: 49.4°C and minimum temperature: 30°C.
- b) Magnesium oxide nanoparticle: Maximum temperature: 46.5°C and minimum temperature: 30°C.
- c) Aluminium oxide nanoparticle: Maximum temperature: 48°C and minimum temperature: 30°C.
- d) Copper oxide nanoparticle: Maximum temperature: 54.4°C and minimum temperature: 30°C.

From the analysis of figure 4 it is clear that the solar heat gain in copper oxide is better than the other nanoparticles which have been used in the present work.

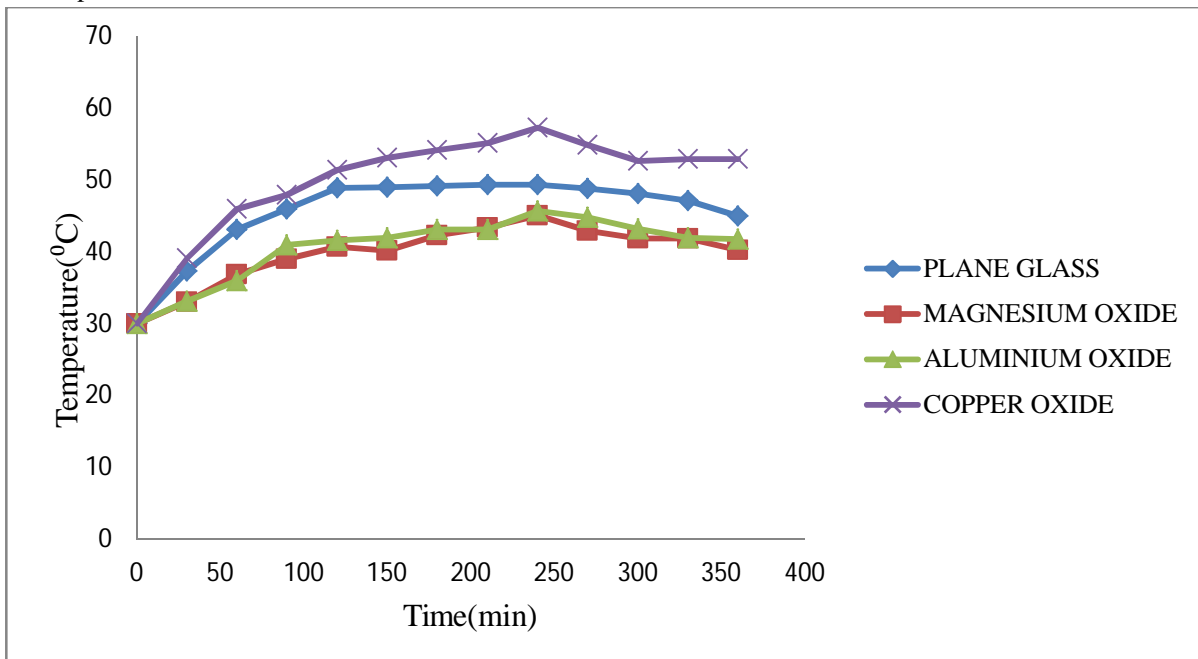


Figure 5 Comparison between time and temperature for 10 gm nanoparticles

When 10gm nanoparticles are mixed with adhesive and coated on the glass faces-

- a) Plane glass: Maximum temperature: 49.4C and minimum temperature: 30⁰C.
- b) Magnesium oxide nanoparticle: Maximum temperature: 44.7 ⁰C and minimum temperature: 30⁰C.
- c) Aluminium oxide nanoparticle: Maximum temperature: 46.5⁰C and minimum temperature: 30⁰C.
- d) Copper oxide nanoparticle: Maximum temperature: 55.9⁰C and minimum temperature: 30⁰C.

From the analysis of figure 5 it is clear that the solar heat gain in copper oxide is better than the other nanoparticles which have been used in the present work.

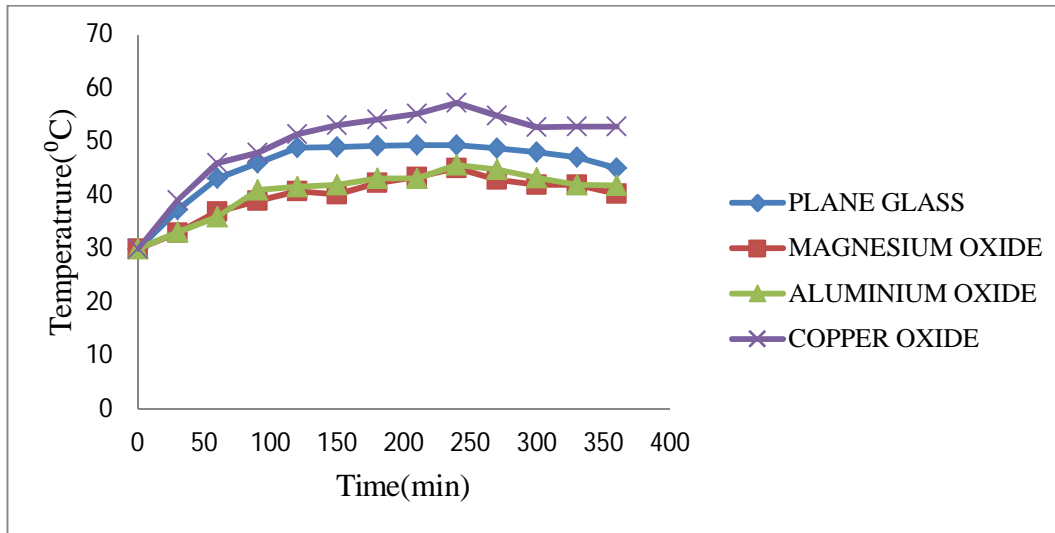


Figure 6 Comparison between time and temperature for 15gm nanoparticles

When 15gm nanoparticles are mixed with adhesive and coated on the glass faces-

- a) Plane glass: Maximum temperature: 49.4C and minimum temperature: 30⁰C.
- b) Magnesium oxide nanoparticle: Maximum temperature: 45.1⁰C and minimum temperature: 30⁰C.
- c) Aluminium oxide nanoparticle: Maximum temperature: 45.7⁰C and minimum temperature: 30⁰C.
- d) Copper oxide nanoparticle: Maximum temperature: 57.3⁰C and minimum temperature: 30⁰C.

From the analysis of figure 6 it is clear that the solar heat gain in copper oxide is better than the other nanoparticles which have been used in the present work.

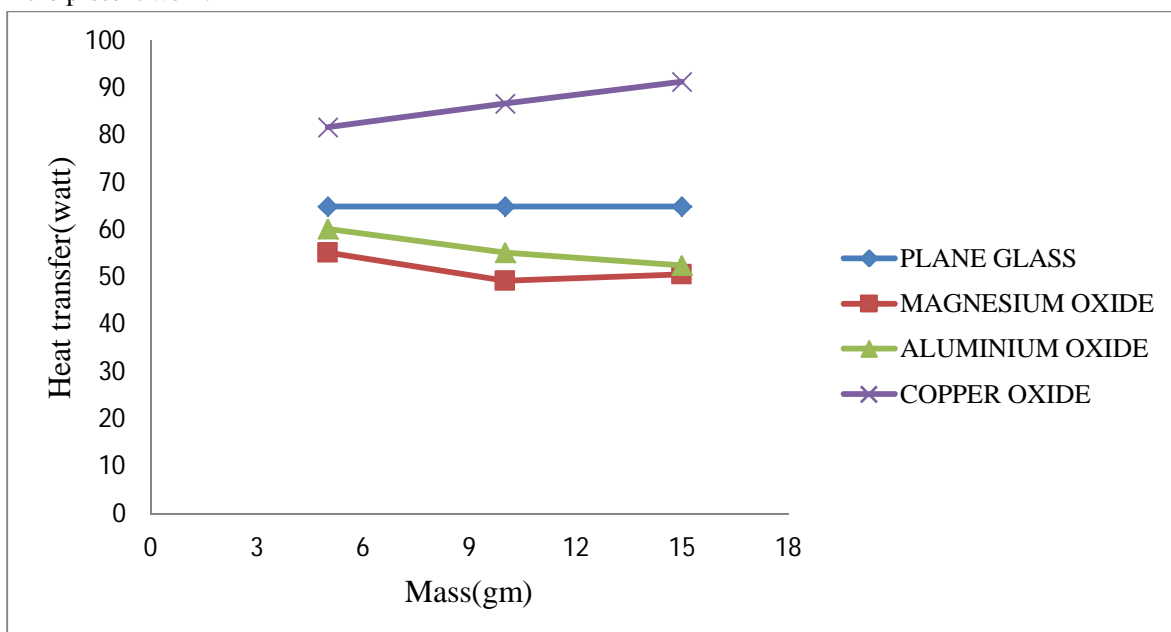


Figure 7 Comparison between solar heat gains in different test zones of different nanoparticles

From the analysis of figure 7 it is clear that the solar heat gain in copper oxide when all the nanoparticles are applied on the ratio of 5gm, 10gm and 15gm is better than the other nanoparticles which have been used in the present work.

IV. CONCLUSION

This type of method is useful for maintaining the high room temperature (comfort temperature) for the places where outside temperature is very low throughout the year. The advantage of this method is that it is low cost exercise and it does not cause any harm to the environment. Here magnesium oxide nanoparticle, aluminium oxide and copper oxide nanoparticle are used for increasing the inside temperature in the test zone. It has been found that copper oxide nanoparticles applied test sections are better for maintaining comfort temperature.

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APPENDIX:

Table 2 Temperature reading when 5gm nanoparticles are mixed with adhesive

TIME	PLANE GLASS	MAGNESIUM OXIDE	ALUMINIUM OXIDE	COPPER OXIDE
0min	30 ⁰ C	30 ⁰ C	30 ⁰ C	30 ⁰ C
30min	37.3 ⁰ C	33.9 ⁰ C	33.9 ⁰ C	38.1 ⁰ C
60min	43.1 ⁰ C	38 ⁰ C	37 ⁰ C	44.8 ⁰ C
90min	46 ⁰ C	40.9 ⁰ C	41.1 ⁰ C	47 ⁰ C

120min	48.9 ⁰ C	41.9 ⁰ C	42.1 ⁰ C	49.1 ⁰ C
150min	49 ⁰ C	42.7 ⁰ C	42.1 ⁰ C	51.9 ⁰ C
180min	49.2 ⁰ C	44.1 ⁰ C	44.5 ⁰ C	54 ⁰ C
210min	49.4 ⁰ C	46.5 ⁰ C	46.3 ⁰ C	54.4 ⁰ C
240min	49.4 ⁰ C	45.1 ⁰ C	48 ⁰ C	52.8 ⁰ C
270min	48.8 ⁰ C	44.3 ⁰ C	47.3 ⁰ C	51 ⁰ C
300min	48.1 ⁰ C	42.1 ⁰ C	44.8 ⁰ C	50.3 ⁰ C
330min	47.1 ⁰ C	41.9 ⁰ C	43.5 ⁰ C	50.3 ⁰ C

Table 3
Temperature reading when 10 gm nanoparticles are mixed with adhesive

TIME	45 ⁰ C PLANE GLASS	MAGNESIUM OXIDE	ALUMINIUM OXIDE	51.1 ⁰ C COPPER OXIDE
0min	30 ⁰ C	30 ⁰ C	30 ⁰ C	30 ⁰ C
30min	37.3 ⁰ C	33.1 ⁰ C	33.3 ⁰ C	38.8 ⁰ C
60min	43.1 ⁰ C	37.1 ⁰ C	36.7 ⁰ C	45.2 ⁰ C
90min	46 ⁰ C	41 ⁰ C	40 ⁰ C	47.1 ⁰ C
120min	48.9 ⁰ C	41.5 ⁰ C	41.7 ⁰ C	49.9 ⁰ C
150min	49 ⁰ C	41.7 ⁰ C	42.1 ⁰ C	52.7 ⁰ C
180min	49.2 ⁰ C	44.3 ⁰ C	42.1 ⁰ C	52.9 ⁰ C
210min	49.4 ⁰ C	44.7 ⁰ C	44 ⁰ C	53 ⁰ C
240min	49.4 ⁰ C	44.7 ⁰ C	46.5 ⁰ C	55.1 ⁰ C
270min	48.8 ⁰ C	42.2 ⁰ C	44.8 ⁰ C	55.9 ⁰ C

Table 4
Temperature reading when 15 gm nanoparticles are mixed with adhesive

TIME	PLANE GLASS	MAGNESIUM OXIDE	ALUMINIUM OXIDE	52.1 ⁰ C COPPER OXIDE
330min	47.1 ⁰ C	41.1 ⁰ C	42.1 ⁰ C	52.1 ⁰ C
0min	30 ⁰ C	30 ⁰ C	30 ⁰ C	30 ⁰ C
360min	45 ⁰ C	40.9 ⁰ C	42.1 ⁰ C	52 ⁰ C
30min	37.3 ⁰ C	33 ⁰ C	33.1 ⁰ C	39.1 ⁰ C
60min	43.1 ⁰ C	36.9 ⁰ C	36 ⁰ C	46 ⁰ C
90min	46 ⁰ C	39 ⁰ C	41 ⁰ C	47.9 ⁰ C
120min	48.9 ⁰ C	40.7 ⁰ C	41.6 ⁰ C	51.4 ⁰ C
150min	49 ⁰ C	40.2 ⁰ C	42 ⁰ C	53.1 ⁰ C
180min	49.2 ⁰ C	42.3 ⁰ C	43.1 ⁰ C	54.2 ⁰ C



210min	49.4 ⁰ C	43.4 ⁰ C	43.1 ⁰ C	55.2 ⁰ C
240min	49.4 ⁰ C	45.1 ⁰ C	45.7 ⁰ C	57.3 ⁰ C
270min	48.8 ⁰ C	42.9 ⁰ C	44.8 ⁰ C	54.9 ⁰ C
300min	48.1 ⁰ C	41.9 ⁰ C	43.2 ⁰ C	52.7 ⁰ C
330min	47.1 ⁰ C	41.9 ⁰ C	42 ⁰ C	52.9 ⁰ C
360min	45 ⁰ C	40.3 ⁰ C	41.8 ⁰ C	52.9 ⁰ C

Table 5 Heat

NANOP
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